

Experiment 1

1. Write the general equation for an equilibrium/phase transition and give the conditions of each. Example: melting

Experiment 2: Liquid and Vapor Equilibrium in a Closed System

In a closed system, equilibrium is established between the liquid and the vapor of a substance. The conditions of equilibrium are:

- a. pressure is equal to the external pressure
b. the temperature is equal to the boiling point

a. $p_{\text{liquid}} = p_{\text{external}}$

b. $T_{\text{liquid}} = T_{\text{boiling}}$

a. $p_{\text{vapor}} = p_{\text{external}}$

b. $T_{\text{vapor}} = T_{\text{boiling}}$

Experiment 3: Vapor Pressure of Liquids

1. Write the general equation for the vapor pressure of a liquid and give the conditions of equilibrium. Example: boiling

a. Write the general equation for the vapor pressure of a liquid and give the conditions of equilibrium.

b. Write the general equation for the vapor pressure of a liquid and give the conditions of equilibrium.

c. Write the general equation for the vapor pressure of a liquid and give the conditions of equilibrium.

d. Write the general equation for the vapor pressure of a liquid and give the conditions of equilibrium.

2. Write the general equation for the vapor pressure of a liquid and give the conditions of equilibrium. Example: boiling

a. $p_{\text{liquid}} = p_{\text{external}}$

b. $T_{\text{liquid}} = T_{\text{boiling}}$

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Arturo Cuomo



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