

Unit 10



UNIT 10: PERIODIC TABLE: CONCEPTS AND APPLICATIONS

Section 10.1: Introduction

In your notebook, read about the modern periodic table.

Copy the information in the box on the left below from the periodic table to complete the table on the right.

Atomic Mass	_____
Atomic Number	_____
Electronic Configuration	_____
Chemical Name	_____
Chemical Symbol	_____

Put each item in Column A, under the letter of the matching item in Column B.

Column A

- _____ 10. A column in the periodic table
- _____ 10. A row in the periodic table
- _____ 20. Group 1 elements
- _____ 20. Elements that are strong oxidising agents
- _____ 20. Group 18 elements

Column B

- a. metals
- b. group
- c. period
- d. representative elements
- e. noble gases

In the space on the left, write each of the statements in terms of the statement in bold, change the underlined word or phrase as needed to fit.

- _____ 20. There are two representative elements in the periodic table.
- _____ 20. There are two representative elements in the periodic table.
- _____ 20. Group 1 elements are highly reactive metals.
- _____ 20. Group 18 elements are the noble gases.
- _____ 20. Group 1 elements are highly reactive metals.
- _____ 20. Group 18 elements are the noble gases.
- _____ 20. Group 1 elements are highly reactive metals.
- _____ 20. Group 18 elements are the noble gases.
- _____ 20. Group 1 elements are highly reactive metals.
- _____ 20. Group 18 elements are the noble gases.

Hydrocarbons Study Guide Answers For Content Mastery

Clifford Lane



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