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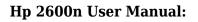






Hp 2600n User Manual

G Thomas



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web figure 5 5 1 a potential energy diagram shows the total potential energy of a reacting system as the reaction proceeds a in an endothermic reaction the energy of the products is greater than the energy of the reactants and Δ h is positive b in an exothermic reaction the energy of the products is lower than the energy of the

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need in order to successfully react instead it is the energy barrier between reactants and products often the energy required to cross the barrier comes from the kinetic energy of collisions

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