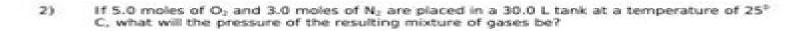




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1)	How many moles of gas occupy 98 L at a pressure of 2.8 atmospheres and a temperature of 292 K?



3) A balloon is filled with 35.0 L of helium in the morning when the temperature is 20.0° C. By noon the temperature has risen to 45.0° C. What is the new volume of the balloon?

4) A 35 L tank of oxygen is at 315 K with an internal pressure of 190 atmospheres. How many moles of gas does the tank contain?

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attraction between the particles is the volume excluded by a mole of particles A careful distinction *Expanding on the Gas Laws*, 2009-01-01 Inquiries in Science Chemistry Series Expanding on the Gas Laws Teacher's Guide **Lesson 45**ADvTECH., <u>Williams & Meyers Oil and Gas Law</u> Howard R. Williams, 2010 **Gas Law Deviation Coefficients for Some Common Gases and Gas Mixtures** William Jasper Clark, Barry Gordon White, 1959 <u>Oil and Gas Law</u>, 1967

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