

Substrate's Degree of Substitution

Based on the substituent groups of the substrate, the degree of substitution of the substrate is called **degree of substitution**. An aromatic compound containing substituents is called **substituted benzene**. There are several types of substituents, which depend on the groups.

- Electrophile** - a substituent which increases the reactivity of the substrate in electrophilic aromatic substitution reaction.
- Nucleophile** - a substituent which decreases the reactivity of the substrate in electrophilic aromatic substitution reaction.
- Ortho-directing** - a substituent which directs the electrophile to the ortho position of the substrate.
- Para-directing** - a substituent which directs the electrophile to the para position of the substrate.

Substituents that push electrons from the aromatic ring are called **electron-releasing groups**. Substituents that pull electrons from the aromatic ring are called **electron-withdrawing groups**. The substituents that push electrons from the aromatic ring are called **electron-releasing groups**.

1. Which of the following is not an electron-releasing group? (1 mark)
 - a. Methyl group
 - b. Hydroxyl group
2. Which of the following is not an electron-releasing group? (1 mark)
 - a. Methyl group
 - b. Hydroxyl group
3. Which of the following is not an electron-releasing group? (1 mark)
 - a. Methyl group
 - b. Hydroxyl group
4. Which of the following is not an electron-releasing group? (1 mark)
 - a. Methyl group
 - b. Hydroxyl group
5. Which of the following is not an electron-releasing group? (1 mark)
 - a. Methyl group
 - b. Hydroxyl group
6. Which of the following is not an electron-releasing group? (1 mark)
 - a. Methyl group
 - b. Hydroxyl group
7. Which of the following is not an electron-releasing group? (1 mark)
 - a. Methyl group
 - b. Hydroxyl group
8. Which of the following is not an electron-releasing group? (1 mark)
 - a. Methyl group
 - b. Hydroxyl group
9. Which of the following is not an electron-releasing group? (1 mark)
 - a. Methyl group
 - b. Hydroxyl group

Macbeth Act Two Standards Focus Figurative Language Answer Key

Annelies Wilder-Smith



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