Ideal vs Combined

3 tricks to tell them apart

	***	make all
Ideal	(PV =	nRT)

Combined $(P_iV/T_i = P_iV/T_i)$

- We will give you 3 things
 - 1P
 - 1V
 - 1T
 - 1 mole or gram (n)
 - (one will be missing)

- We will give you 5 things
 - 2 P's
 - 2 V's
 - 275
 - (one will be missing)

- 2. Moles or grams will be mentioned mole, part aton
- 2. Moles or grams will not be mentioned

3. The gas will not change

- The gas will change - "gas is heated/cool
- "gas is compressed,
- "gas is pressurized/

Gas Laws Test Review Guide

S Ben Porath

Gas Laws Test Review Guide:

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