

Gas Laws Lab

Instructions: Complete all sections of this report and answer all questions in complete sentences for full credit.

Title:

Objective(s):

Hypothesis: *Write a hypothesis that relates temperature and volume of a gas.*

The hotter the temperature gets, the more volume the gas will take up in the syringe.

Materials: What scientific equipment and tools were used to collect data in the simulation?
(Record the materials as you work through the simulation.)

| Materials |
|---|
| 10-mL Syringe (2) Syringe Tip Cap 5 Water Baths (ice & saltwater, ice water, room temp water, warm water, and hot water). Hydrogen H_2 Helium He Nitrogen N_2 Beaker Thermometer |

Variables: *List variables below as you work through the simulation.*

| Type | Variable |
|-------------|--|
| Controlled | The syringes, the amount of air put into the syringe (tried to keep the same), the air used for the first syringe, the air used for the second syringe (hydrogen, helium, or nitrogen), the beakers all held the same amount of water, the thermometer was the same. |
| Independent | The temperature of the water |
| Dependent | The volume of air in each trial, that being the normal air and the hydrogen gas used. |

Gas Laws Lab Answers

Jessica J Manson

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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