Second Semester Final Exam

Multiple Choice: Please mark the answer to the multiple choice questions on your bubble sheet.

- 1) The molar mass of magnesium cyanide, Mg(CN)2, is approximately
 - A) 50.3 g
 - B) 64.3 g
 - C) 76.3 g
 - D) 100.6 g
- 2) The total number of atoms in (NH₄)₃BO₃ is
 - A) 12
 - B) 17
 - C) 19
- 3) What is the mass in grams of 5.90 mol C₈H₁₈?
 - A) 0.0512 g
 - B) 389 g
 - C) 673 g
 - D) 19.4 g
- 4) How many atoms are there in 1 mole of helium?
 - A) 6.02 x 10²³ atoms
 - B) 22.4 atoms
 - C) 4.0 atoms
 - D) 0.0821 atoms
- 5) The percent composition of silver (Ag) in AgNO3 is
 - A) 0.625 %
 - B) 1.57 %
 - C) 62.5%
 - D) 157 %
 - E) 169.9 %
- 6) Which of the following is an empirical formula?
 - A) C₂H₄
 - B) C₂H₆
 - C) CH4
 - D) C₃H₆
- Which of the following could be a molecular formula for the empirical formula CH2
 - A) C₃H₆
 - B) C₆H₃
 - C) CH
 - D) CH4
- 8) What is the coefficient in front of the O2 when the following reaction is correctly balanced?

 $_{C_3H_8} + _{CO_2} \rightarrow _{CO_2} + _{H_2O}$

- A) 1 B) 2
 - 2 C) 3
- D) 4
- E) 5

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Michael Seilmaier

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