




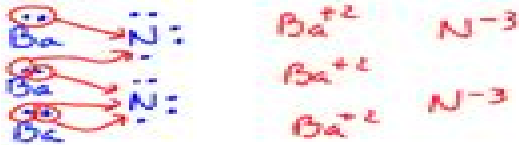








Draw the dot diagrams for each element in the compound

<p>1. NaCl</p> 	<p>7. Li<sub>2</sub>O</p> 
<p>2. CaO</p> 	<p>8. Ag<sub>3</sub>P</p> 
<p>3. AlN</p> 	<p>9. Ba<sub>3</sub>N<sub>2</sub></p> 
<p>4. MgS</p> 	<p>10. Al<sub>2</sub>O<sub>3</sub></p> 
<p>5. FeCl<sub>2</sub></p> 	<p>11. ZnI<sub>2</sub></p> 
<p>6. CoF<sub>3</sub></p> 	<p>12. Au<sub>2</sub>S<sub>3</sub></p> 

# Modeling Lewis Dot Structures Answer Key

**VM Jensen**



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