

Chemistry – Unit 7 Review

Chemical Reaction Model

1. Describe key characteristics of all chemical reactions, including the role of energy. Explain how a balanced equation represents these features (include an example).

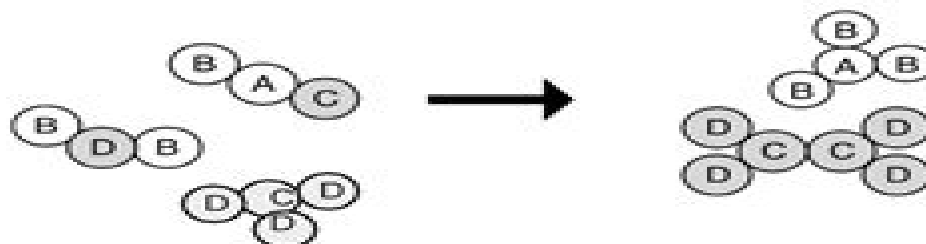
In chemical reactions, atoms of the reactants recombine to form new substances in the products. There is almost always some transfer between the thermal and chemical energy accounts as well as an exchange of thermal energy via heating (Q) between the system and surroundings. The balanced chemical equation represents this process symbolically, showing that the atoms in the products are the same as those in the reactants



Note: there are 4 atoms of H and 2 atoms of O on each side of the equation.

Applying the Model

2. Balance the reaction below by drawing any additional reactant and/or product molecules that would be needed.



Write the **balanced equation** for this reaction.



Which of these molecules would be called reactants?

DB₂, CD₂, and BAC are reactants

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Annelies Wilder-Smith



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Matriculation Model Answers: Chemistry University Correspondence College,1922 Modeling of Molecular Properties Peter Comba,2011-10-05 Molecular modeling encompasses applied theoretical approaches and computational techniques to model structures and properties of molecular compounds and materials in order to predict and or interpret their properties The modeling covered in this book ranges from methods for small chemical to large biological molecules and materials With its comprehensive coverage of important research fields in molecular and materials science this is a must have for all organic inorganic and biochemists as well as materials scientists interested in applied theoretical and computational chemistry The 28 chapters written by an international group of experienced theoretically oriented chemists are grouped into four parts Theory and Concepts Applications in Homogeneous Catalysis Applications in Pharmaceutical and Biological Chemistry and Applications in Main Group Organic and Organometallic Chemistry The various chapters include concept papers tutorials and research reports **Model Answers on Chemistry. (General Certificate of Education.).** Elizabeth Joan Perkins,1967 Using Modeling to Gain a Better Understanding of Chemistry Concepts Elisabeth Knierim,2018 Molecular Modeling Basics Jan H. Jensen,2017-06-29 Molecular modeling is becoming an increasingly important part of chemical research and education This volume provides the fundamental theory needed to understand not only what molecular modeling programs do but also the gist of research papers that describe molecular modeling results It begins by examining the potential energy surface PES **Model Set & Guide** Peter Atkins,Loretta Jones,2002-05-17 *Model Answers on Chemistry* Elizabeth Joan Perkins,1967 **Pure Chemistry** ,1975 Model Questions and Answers in Chemistry Marthinus Jacobus Oosthuizen,1970 *Chemistry (Advanced Level)* Louis Alexander Linton,1965

Introducing Molecular Modeling Into the Undergraduate Chemistry Curriculum Warren J. Hehre,Janet E. Nelson,1997 Chemistry Model Answers RND Publications Ltd,1998-12-01 **Chemistry for Senior Students** Charles Alfred Janson,1986 **Advanced Level Chemistry** Vijay Pandit,1990-04-01

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