

## Chemistry – Unit 4 Worksheet 4

**DMC.5:** *I can cite and interpret the evidence that supports the belief that some pure substances are made up of simpler particles (atoms) in a definite ratio.*

Answer the following questions. Be sure to show all mathematical work and reasoning and use complete sentences in explanations.

- Two compounds of hydrogen and oxygen are tested. Compound I contains 120.0 g of oxygen and 15.0 g of hydrogen. Compound II contains 32.0 g of oxygen and 2.0 g of hydrogen.
  - Determine the ratio of the mass of oxygen to the mass of hydrogen in each of the compounds.
  - Why are the compounds not the same?
  - What is significant about these mass ratios?
  - If compound I is water, what could be the formula of compound II?
- Nitrogen and oxygen combine to form a variety of compounds. The following data were collected for three different compounds of nitrogen and oxygen:

Analysis Data of Nitrogen & Oxygen Compounds	
Compound	Mass of Nitrogen that combines with every 1.0 g of Oxygen
A	1.750 g
B	0.8750 g
C	0.4375 g

- Additional evidence shows that the formula of compound B is NO. Sketch particle diagrams of molecules of all three compounds.
- Justify your representations above.
- There is no compound between Nitrogen and oxygen with a mass ratio of 2.00 grams of N for every 1.00 grams of oxygen. Why?

# Modeling Chemistry Utest Answers

**Michael Brown**



## **Modeling Chemistry Utest Answers:**

## Decoding **Modeling Chemistry Utest Answers**: Revealing the Captivating Potential of Verbal Expression

In an era characterized by interconnectedness and an insatiable thirst for knowledge, the captivating potential of verbal expression has emerged as a formidable force. Its ability to evoke sentiments, stimulate introspection, and incite profound transformations is genuinely awe-inspiring. Within the pages of "**Modeling Chemistry Utest Answers**," a mesmerizing literary creation penned by a celebrated wordsmith, readers embark on an enlightening odyssey, unraveling the intricate significance of language and its enduring effect on our lives. In this appraisal, we shall explore the book's central themes, evaluate its distinctive writing style, and gauge its pervasive influence on the hearts and minds of its readership.

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