

1. Which is not an example of exothermic reaction :



2. Which is an example of endothermic reaction :



3. Which is an example of combination reaction :



4. Which is an example of decomposition reaction?



5. Which is an example of thermal decomposition?



6. Which is an example of electrolysis decomposition?



7. Which is an example of photo chemical reactions?



8. Which compound is used in photography?



9. Which is an example of double displacement reaction?



10. $\text{Fe}_2\text{O}_3 + 2\text{Al} \rightarrow \text{Al}_2\text{O}_3 + 2\text{Fe}$ is an example of

(a) Combination reaction

(c) Double displacement reaction

11. Which is an example of double displacement reaction?



12. $\text{H}_2\text{S} + \text{Cl}_2 \rightarrow 2\text{HCl} + \text{S}$ is this reaction 'Cl₂' act as :

(a) Oxidising agent

(b) Reducing agent

13. Which statement is correct regarding the reaction $2\text{PbO} + \text{C} \rightarrow \text{C}_2\text{Pb} + \text{CO}_2$

(a) Lead is getting reduced (b) Carbon dioxide is getting oxidised

(c) Lead oxide is getting reduced (d) All of these

14. In the reaction $\text{MnO}_2 + 4\text{HCl} \rightarrow \text{MnCl}_2 + 2\text{H}_2\text{O} + \text{Cl}_2$, MnO_2 act as?

(a) Reducing agent

(b) Oxidising agent

15. Which is an example of redox reaction?



16. Which is a formula of rust?



17. Rusting of Iron is an example:

(a) Oxidation

(b) Reduction

18. Antioxidant is/are:

(a) Oxidising agent

(b) Reducing agent

19. Rancidity can be prevented by:

(a) Adding antioxidants



(d) All of these



(d) All of these



(d) All of these



(d) All of these



(d) All of these



(d) (a) and (b)



(d) All of these

(b) Decomposition reaction

(d) Displacement reaction



(c) Both (a) & (b)

(d) None of these

(c) Both (a) & (b)

(d) None of these



(d) All of these



(c) Redox

(d) All of these

(c) Both (a) & (b)

(d) None of these

(b) Storing food in air light

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Roman Wölfel

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Table of Contents Modeling Chemistry Unit 10 Ws 2 Answers

1. Understanding the eBook Modeling Chemistry Unit 10 Ws 2 Answers
 - The Rise of Digital Reading Modeling Chemistry Unit 10 Ws 2 Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Modeling Chemistry Unit 10 Ws 2 Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Modeling Chemistry Unit 10 Ws 2 Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Modeling Chemistry Unit 10 Ws 2 Answers
 - Personalized Recommendations

- Modeling Chemistry Unit 10 Ws 2 Answers User Reviews and Ratings
- Modeling Chemistry Unit 10 Ws 2 Answers and Bestseller Lists
- 5. Accessing Modeling Chemistry Unit 10 Ws 2 Answers Free and Paid eBooks
 - Modeling Chemistry Unit 10 Ws 2 Answers Public Domain eBooks
 - Modeling Chemistry Unit 10 Ws 2 Answers eBook Subscription Services
 - Modeling Chemistry Unit 10 Ws 2 Answers Budget-Friendly Options
- 6. Navigating Modeling Chemistry Unit 10 Ws 2 Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - Modeling Chemistry Unit 10 Ws 2 Answers Compatibility with Devices
 - Modeling Chemistry Unit 10 Ws 2 Answers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Modeling Chemistry Unit 10 Ws 2 Answers
 - Highlighting and Note-Taking Modeling Chemistry Unit 10 Ws 2 Answers
 - Interactive Elements Modeling Chemistry Unit 10 Ws 2 Answers
- 8. Staying Engaged with Modeling Chemistry Unit 10 Ws 2 Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Modeling Chemistry Unit 10 Ws 2 Answers
- 9. Balancing eBooks and Physical Books Modeling Chemistry Unit 10 Ws 2 Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Modeling Chemistry Unit 10 Ws 2 Answers
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Modeling Chemistry Unit 10 Ws 2 Answers
 - Setting Reading Goals Modeling Chemistry Unit 10 Ws 2 Answers
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Modeling Chemistry Unit 10 Ws 2 Answers
 - Fact-Checking eBook Content of Modeling Chemistry Unit 10 Ws 2 Answers

- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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