Name	
Date	Per

Mixed Practice: Names and Formulas

First determine whether each of the following compounds is *ionic* (M + NM), polyatomic (3 or more atoms), or covalent (2 NM's). Then write the name for each compound.

1)	NO ₂
	NaBr
	SiO ₂
4)	P ₂ Br ₄
5)	FeSO ₄
6)	SF ₆
7)	Li ₂ S
8)	MgBr ₂
9)	N ₂ S
10) Be(OH) ₂
11) SO ₃
12) Cu ₂ S
13) BF ₃
14) I ₂
15) Pb ₃ (PO ₃) ₂

Modeling Chemistry Unit 4 Answers

Thirumalaisamy P. Velavancorresponding

Modeling Chemistry Unit 4 Answers:

This book delves into Modeling Chemistry Unit 4 Answers. Modeling Chemistry Unit 4 Answers is an essential topic that must be grasped by everyone, ranging from students and scholars to the general public. The book will furnish comprehensive and in-depth insights into Modeling Chemistry Unit 4 Answers, encompassing both the fundamentals and more intricate discussions.

- 1. This book is structured into several chapters, namely:
 - Chapter 1: Introduction to Modeling Chemistry Unit 4 Answers
 - Chapter 2: Essential Elements of Modeling Chemistry Unit 4 Answers
 - o Chapter 3: Modeling Chemistry Unit 4 Answers in Everyday Life
 - Chapter 4: Modeling Chemistry Unit 4 Answers in Specific Contexts
 - ∘ Chapter 5: Conclusion
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- 3. In chapter 2, this book will delve into the foundational concepts of Modeling Chemistry Unit 4 Answers. This chapter will elucidate the essential principles that must be understood to grasp Modeling Chemistry Unit 4 Answers in its entirety.
- 4. In chapter 3, the author will examine the practical applications of Modeling Chemistry Unit 4 Answers in daily life. The third chapter will showcase real-world examples of how Modeling Chemistry Unit 4 Answers can be effectively utilized in everyday scenarios.
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- 6. In chapter 5, this book will draw a conclusion about Modeling Chemistry Unit 4 Answers. This chapter will summarize the key points that have been discussed throughout the book.
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