Properties of Ionic Compounds

Context: Molecules with ionic bonds have many similar properties that occur because of the properties of ionic bonds as well as the crystalline structure of the ionic solids. Because of these common properties of ionic compounds it is important to identify why these properties happen by examining the molecules at the molecular level.

Model I: Ionic Compound Demonstration Data Table

	NaCl Table Salt	NaHCO ₃ Baking Soda	MgSO, Epoom Salt	CaCO ₃ Limestone /Antacid
Britise?				
Conducts Electricity as a solid?				
Wister Soluble?				
Conducts Electricity in water?				
Largest. Esectronegativity Difference				

Questions:

I) Identify which two atoms in each molecule would have the largest electronegativity difference. Record this on the data table.

Ionic Covalent Properties Pogil

ML Morrison

Ionic Covalent Properties Pogil:

<u>Process Oriented Guided Inquiry Learning (POGIL)</u> Richard Samuel Moog,2008 POGIL is a student centered group learning pedagogy based on current learning theory This volume describes POGIL s theoretical basis its implementations in diverse environments and evaluation of student outcomes

**Russian Journal of Coordination Chemistry*, 1998*

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