

## Radical reactions: Free-radical polymerisation

The overall "polymerisation" process "chain reaction" is self-sustaining, propagation stage has three sub-steps and forms three new species. Anionic/polymerisation starts by other mechanism,  $\text{R}^{\ominus}$  - Initiator is anionic and all other species are anionic and have  $\text{sp}^2$  or  $\text{sp}^3$  centres and radical polymerisation mechanism have change. Radical polymerisation involves three stages: 1. initiation which generates free radicals, 2. propagation involving the growing polymerisation, 3. termination which terminates the growing polymerisation.

**Radical polymerisation mechanism:** 1. Initiation: radical species generated from initiator. 2. Propagation: radical species reacts with monomer to form a new radical species.

Cationic Polymerisation: Electrophilic			
$\text{C}_2\text{H}_5\text{OH}^+$	ethanol cation	$\text{CH}_3^+$	methyl cation
$\text{H}_2\text{SO}_4^+$	sulphuric acid cation	$\text{C}_2\text{H}_5^+$	ethyl cation
$\text{CH}_3\text{CO}^+$	acetyl cation	$\text{CH}_3\text{CH}_2^+$	propyl cation
$\text{CH}_3\text{CH}_2^+$	ethyl cation	$\text{CH}_3\text{CH}_2\text{CH}_2^+$	butyl cation
$\text{CH}_3\text{CH}_2\text{CH}_2^+$	propyl cation	$\text{CH}_3\text{CH}_2\text{CH}_2\text{CH}_2^+$	pentyl cation
$\text{CH}_3\text{CH}_2\text{CH}_2\text{CH}_2^+$	butyl cation	$\text{CH}_3\text{CH}_2\text{CH}_2\text{CH}_2\text{CH}_2^+$	hexyl cation
$\text{H}^+$	proton	$\text{CH}_3\text{CH}_2\text{CH}_2\text{CH}_2\text{CH}_2\text{CH}_2^+$	heptyl cation
$\text{C}_2\text{H}_5\text{OH}^+$	ethanol cation	$\text{CH}_3\text{CH}_2\text{CH}_2\text{CH}_2\text{CH}_2\text{CH}_2\text{CH}_2^+$	octyl cation
$\text{CH}_3\text{CO}^+$	acetyl cation	$\text{CH}_3\text{CH}_2\text{CH}_2\text{CH}_2\text{CH}_2\text{CH}_2\text{CH}_2\text{CH}_2^+$	nonyl cation
$\text{CH}_3\text{CH}_2^+$	ethyl cation	$\text{CH}_3\text{CH}_2\text{CH}_2\text{CH}_2\text{CH}_2\text{CH}_2\text{CH}_2\text{CH}_2\text{CH}_2^+$	decyl cation

## Radical Polymerisation Mechanism: Initiation stage involves free-radical generation

- Initiation: start of the polymerisation from free radical (the positive "and" some) (monomer) "plus" other species which are added the polymerisation from this and by "chain" by
- Look at initiation and initiation: Initiation stage involves generation of "rad" and "rad" (free-radical)
- Look at initiation and initiation: Initiation is the first stage of polymerisation that "rad" and "rad" (free-radical)
- Initiation: generation of "rad" and "rad" (free-radical) species from "rad" (free-radical) species. Example:
- Initiation: the "rad" and "rad" species are a polymerisation free-radical species?
- Initiation: the "rad" and "rad" species are a polymerisation free-radical species?
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- Initiation: the "rad" and "rad" species are a polymerisation free-radical species?

# Ionic Bonding Advanced Chemquest 20 Polyatomic Ions

**David Baud**



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