

IONIC AND METALLIC BONDING

Homework

Set 1: IONS

- For each element below, state (i) the number of valence electrons in the atom, (ii) the electron dot structure, and (iii) the chemical symbol(s) for the most stable ion.
 - Ba
 - I
 - K
- How many valence electrons does each of the following atoms have?
 - gallium
 - fluorine
 - selenium
- Write the electron configuration for each of the following atoms and ions.
 - Ca
 - chlorine atom
 - Na^+
 - phosphide ion
 - O^{2-}
- What is the relationship between the group number of the representative elements and the number of valence electrons?
- How many electrons will each element gain or lose in forming an ion? State whether the resulting ion is a cation or an anion.
 - strontium
 - aluminum
 - tellurium
 - rubidium
 - bromine
 - phosphorus
- Give the name and symbol of the ion formed when
 - a chlorine atom gains one electron.
 - a potassium atom loses one electron.
 - an oxygen atom gains two electrons.
 - a barium atom loses two electrons.
- How many electrons are lost or gained in forming each of the following ions?
 - Mg^{2+}
 - Br^-
 - Ag^+
 - Fe^{3+}
- Classify each of the following as a cation or an anion.
 - Na^+
 - Cu^{2+}
 - I^-
 - O^{2-}
 - Ca^{2+}
 - Cs^+

Set 2: IONIC BONDS AND IONIC COMPOUNDS

- Use electron dot structures to predict the formula of the ionic compounds formed when the following elements combine.
 - sodium and bromine
 - sodium and sulfur
 - calcium and iodine
 - aluminum and oxygen
 - barium and chlorine
- Which of these combinations of elements are most likely to react to form ionic compounds?
 - sodium and magnesium
 - barium and sulfur
 - potassium and iodine
 - oxygen and argon
- What is the meaning of coordination number?
- How is the coordination number determined?

Ionic And Metallic Bonding Test A Answers

Arturo Cuomo



Ionic And Metallic Bonding Test A Answers:

The Top Books of the Year Ionic And Metallic Bonding Test A Answers The year 2023 has witnessed a remarkable surge in literary brilliance, with numerous engrossing novels captivating the hearts of readers worldwide. Lets delve into the realm of top-selling books, exploring the fascinating narratives that have enthralled audiences this year. Ionic And Metallic Bonding Test A Answers : Colleen Hoovers "It Ends with Us" This heartfelt tale of love, loss, and resilience has captivated readers with its raw and emotional exploration of domestic abuse. Hoover skillfully weaves a story of hope and healing, reminding us that even in the darkest of times, the human spirit can triumph. Uncover the Best : Taylor Jenkins Reids "The Seven Husbands of Evelyn Hugo" This spellbinding historical fiction novel unravels the life of Evelyn Hugo, a Hollywood icon who defies expectations and societal norms to pursue her dreams. Reids absorbing storytelling and compelling characters transport readers to a bygone era, immersing them in a world of glamour, ambition, and self-discovery. Discover the Magic : Delia Owens "Where the Crawdads Sing" This captivating coming-of-age story follows Kya Clark, a young woman who grows up alone in the marshes of North Carolina. Owens spins a tale of resilience, survival, and the transformative power of nature, entrancing readers with its evocative prose and mesmerizing setting. These top-selling novels represent just a fraction of the literary treasures that have emerged in 2023. Whether you seek tales of romance, adventure, or personal growth, the world of literature offers an abundance of captivating stories waiting to be discovered. The novel begins with Richard Papen, a bright but troubled young man, arriving at Hampden College. Richard is immediately drawn to the group of students who call themselves the Classics Club. The club is led by Henry Winter, a brilliant and charismatic young man. Henry is obsessed with Greek mythology and philosophy, and he quickly draws Richard into his world. The other members of the Classics Club are equally as fascinating. Bunny Corcoran is a wealthy and spoiled young man who is always looking for a good time. Charles Tavis is a quiet and reserved young man who is deeply in love with Henry. Camilla Macaulay is a beautiful and intelligent young woman who is drawn to the power and danger of the Classics Club. The students are all deeply in love with Morrow, and they are willing to do anything to please him. Morrow is a complex and mysterious figure, and he seems to be manipulating the students for his own purposes. As the students become more involved with Morrow, they begin to commit increasingly dangerous acts. The Secret History is a brilliant and thrilling novel that will keep you wondering until the very end. The novel is a cautionary tale about the dangers of obsession and the power of evil.

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