- 2. What does the number next to isotopes signify? | 1'5 +he mass
- 3. How can you tell isotopes apart? by the number of neutrons

For each of the following isotopes, write the number of protons, neutrons, and electrons.

| | Chromium-58 | Chromium-63 |
|----------------|-------------|-------------|
| # of protons | 24 | 2 |
| # of neutrons | | |
| # of electrons | | , |

| | Carbon-12 | Carbon-16 |
|----------------|-----------|-----------|
| # of protons | | |
| # of neutrons | | |
| # of electrons | - | |

| | Nitrogen-15 | Nitrogen-20 |
|----------------|-------------|-------------|
| # of protons | | |
| # of neutrons | | |
| # of electrons | | |

| | Sulfur-23 | Sulfur-25 |
|----------------|-----------|-----------|
| # of protons | | |
| # of neutrons | | _ |
| # of electrons | | + |

Ion And Isotopes Practice Set Answers

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