



### Extended Response

*Use the information below to answer questions 15–18.*

A proton, which has a mass of  $1.673 \times 10^{-27}$  kg, accelerates from rest in a uniform electric field of 640 N/C. At some time later, its speed is  $1.2 \times 10^6$  m/s.

15. What is the magnitude of the acceleration of the proton?

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**Te HS&T** a Holt Rinehart & Winston, Holt, Rinehart and Winston Staff, 2004-02      **Children's Books in Print** R R Bowker Publishing, Bowker, 1999-12      **Achievement Testing in Physics, Selected Reprints** American Association of Physics Teachers, 1965      **Resources in Education**, 1984      **Resource Letters** American Association of Physics Teachers, 1966      **John Stewart Bell and Twentieth-Century Physics** Andrew Whitaker, 2016-07-07 John Stewart Bell 1928 1990 was one of the most important figures in twentieth century physics famous for his work on the fundamental aspects of the century's most important theory quantum mechanics While the debate over quantum theory between the supremely famous physicists Albert Einstein and Niels Bohr appeared to have become sterile in the 1930s Bell was able to revive it and to make crucial advances Bell's Theorem or Bell's Inequalities He was able to demonstrate a contradiction between quantum theory and essential elements of pre quantum theory locality and causality The book gives a non mathematical account of Bell's relatively impoverished upbringing in Belfast and his education It describes his major contributions to quantum theory but also his important work in the physics of accelerators and nuclear and elementary particle physics      **Chemical Principles in Practice** Jerry A. Bell, 1967 This forward looking laboratory text includes thirty three experiments designed to serve high level introductory chemistry courses It is intended in part to give the students practice in techniques not only those generally considered the techniques of chemistry but also those of mathematics However the only prerequisite is a good background in high school mathematics Ideally suited for use with University Chemistry or College Chemistry by Bruce Mahan this book follows the physical chemistry approach of these and most other modern chemistry texts maintaining the same flavor emphasis and degree of sophistication All experiments are open ended and include suggestions for following them up There is considerable emphasis on the involvement of the student in planning his own procedures and quantitative thinking Publisher      *Te HS&T J* Holt Rinehart & Winston, Holt, Rinehart and Winston Staff, 2004-02      **Books in Print Supplement**, 2002      *Resource Letters* American Institute of Physics, 1961

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