

Balls Worksheet

Directed Reading A

Section: Physical Properties

PHYSICAL PROPERTIES

_____ 1. A characteristic of matter that can be observed or measured without changing the identity of the matter is a _____.

- | | |
|------------------------------|-----------------------|
| a. matter property. | c. chemical property. |
| b. physical property. | d. volume property. |

_____ 2. Some examples of physical properties are _____.

- | | |
|----------------------------|---|
| a. color, odor, and age. | b. volume, color, and magnetism. |
| c. color, odor, and speed. | d. volume, color, and age. |

Match the correct example with the correct physical property. Write the letter in the space provided.

- | | |
|---|-------------------------|
| D _____ 3. Aluminum can be flattened into sheets of foil. | a. state |
| A _____ 4. An ice cube floats in a glass of water. | b. solubility |
| F _____ 5. Copper can be pulled into thin wires. | c. thermal conductivity |
| C _____ 6. Plastic foam protects you from hot liquid. | d. malleability |
| B _____ 7. Flavored drink mix dissolves in water. | e. odor |
| E _____ 8. An onion gives off a very distinctive smell. | f. ductility |
| G _____ 9. A gold ball has more mass than a table tennis ball. | g. density |

10. Density is the _____ **Physical Property** _____ that describes the relationship between mass and volume.

11. Objects such as a cotton ball and a small tomato can occupy similar volumes but vary greatly in _____ **Density** _____.

12. If you pour different liquids into a graduated cylinder, the liquids will form layers based upon differences in the _____ **Density** _____ of each liquid.

13. Which layer of liquid would settle on the bottom of a graduated cylinder? **The**

Holt Environmental Science Chapter 9 Review Answers

Yi-Tong Ma



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