

Acid Precipitation (Environmental Science)

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Introduction text

What is Acid Precipitation?

Acid precipitation occurs when rain, snow, or fog that has a low pH (below 5.6) falls to the ground. This precipitation is often acidic, and is caused by the presence of sulfur dioxide and nitrogen oxides in the atmosphere.

Causes of Acid Precipitation

Acid precipitation is caused by the presence of sulfur dioxide and nitrogen oxides in the atmosphere. These pollutants are released by the burning of fossil fuels, such as coal and oil, in power plants and other industrial facilities.

Effects of Acid Precipitation on the Environment

Acid precipitation can have a variety of effects on the environment, including the acidification of lakes and streams, the damage to forests and crops, and the corrosion of buildings and infrastructure.

Effects of Acid Precipitation on Human Health

Acid precipitation can also have effects on human health, such as the irritation of the respiratory system and the damage to the skin. It can also contribute to the acidification of drinking water.

How to Reduce Acid Precipitation

There are several ways to reduce the amount of acid precipitation, including the use of cleaner energy sources, the implementation of emissions trading, and the use of scrubbers to remove pollutants from power plants.

Conclusion

Acid precipitation is a serious environmental problem that can have a variety of effects on the environment and human health. It is caused by the presence of sulfur dioxide and nitrogen oxides in the atmosphere, and can be reduced by the use of cleaner energy sources and the implementation of emissions trading.

Multiple Choice Questions

1. What is the primary cause of acid precipitation?

a. Sulfur dioxide
b. Nitrogen oxides
c. Carbon dioxide
d. Methane

Short Answer Questions

1. How does acid precipitation affect the environment?

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Multiple Choice Questions

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a. Sulfur dioxide
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2. How does acid precipitation affect the environment?

a. It causes the acidification of lakes and streams.
b. It damages forests and crops.
c. It corrodes buildings and infrastructure.
d. All of the above.

3. How can acid precipitation be reduced?

a. By using cleaner energy sources.
b. By implementing emissions trading.
c. By using scrubbers to remove pollutants from power plants.
d. All of the above.

4. What is the pH of acid precipitation?

a. Below 5.6
b. Above 5.6
c. Exactly 5.6
d. None of the above.

5. What is the most common pollutant in acid precipitation?

a. Sulfur dioxide
b. Nitrogen oxides
c. Carbon dioxide
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Answer: a b c d e

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Holt Environmental Science Acid Precipitation Quiz

DP Hallahan



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