

Chemistry Unit 8 Worksheet 4

Samples of Every Kind of Problem

Let us suppose that our "paper" writer is a chemist who has to write out what products may be made. What are the possibilities and how do you do it? The next 4 problems cover these aspects.

1. Chlorine reacts with bromine in a reaction of gaseous molecules. $\text{Br}_2 + \text{Cl}_2$ say that, would also represent the products produced in this reaction. BrCl say, would it be considered a separate gas?

Reactants	1 Br_2	1 Cl_2	0 BrCl
Products	0 Br_2	0 Cl_2	0 BrCl
Change	1 Br_2	1 Cl_2	0 BrCl

Is chlorine
a stable product?

2. How about a displacement reaction, suppose we react with hydrogen bromide solution, we obtain hydrogen gas (which we can break apart now). How many moles of hydrogen bromide will be considered to react with the same mole of H_2 if it is considered?

Reactants	1 H_2	1 HBr	0 H_2	0 HBr
Products	0 H_2	0 HBr	1 H_2	0 HBr
Change	1 H_2	1 HBr	1 H_2	0 HBr

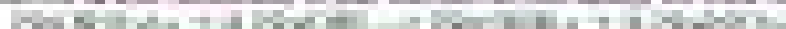
Is hydrogen
a stable product?

3. Chlorine, Cl_2 , reacts with hydrogen gas to produce hydrogen chloride gas (but that is wrong). What would be the separate gas by breaking up into a mole of H_2 and a mole of Cl_2 ?

Reactants	1 Cl_2	1 H_2	0 HCl	0 H_2	0 Cl_2
Products	0 Cl_2	0 H_2	0 HCl	0 H_2	0 Cl_2
Change	1 Cl_2	1 H_2	0 HCl	0 H_2	0 Cl_2

Is hydrogen a stable product? H_2 is a stable product.

4. Chlorine reacts with hydrogen gas to produce hydrogen chloride gas (but that is wrong). What would be the separate gas by breaking up into a mole of H_2 and a mole of Cl_2 ?



What is a stable product?

Reactants	1 H_2	1 Cl_2	0 HCl	0 H_2	0 Cl_2
Products	0 H_2	0 Cl_2	0 HCl	0 H_2	0 Cl_2
Change	1 H_2	1 Cl_2	0 HCl	0 H_2	0 Cl_2

Is hydrogen a stable product? H_2 is a stable product.

Modeling Chemistry U8 Empfrm Practice V2 0 Answers

Ensheng Dong



Modeling Chemistry U8 Empfrm Practice V2 0 Answers:

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