

Review Sheet for Chemistry* First Semester Final

Refer to your class notes, worksheets, and the textbook to complete this review sheet. Study early so that you will have time to ask questions about what you don't understand.

(* The study of matter and the changes it undergoes)

Matter: Anything that takes up space and has mass

➤ Physical Changes and Chemical Changes

Define each. How can you tell the difference between the two?

Classify the following as physical or chemical changes:

- a. spoiling of milk _____
- b. bending wire _____
- c. cutting paper _____
- d. rusting of a nail _____

➤ Put the following into a graphic organizer/flowchart and define each:

- Matter
- Pure substances – elements and compounds
- Mixtures – homogeneous (solutions) and heterogeneous

Identify the following as pure substances, homogeneous mixtures or heterogeneous mixtures:

- a. copper _____
- b. sweetened tea _____
- c. sand and water _____
- d. calcium carbonate (CaCO_3) _____

➤ Sketch particles in the three states of matter. How close are the particles and how much do they move?

Solid	Liquid	Gas

➤ Calculations using the Law of Conservation of Mass for Reactions



Atom

➤ For this Carbon-14 isotope, $^{14}_6\text{C}$

Modeling Chemistry 1st Semester Final V2 3

University of Michigan--Dearborn

A red circular graphic with a gradient, appearing as a partial circle or a stylized 'C' shape, located to the right of the University of Michigan--Dearborn text.

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Table of Contents Modeling Chemistry 1st Semester Final V2 3

1. Understanding the eBook Modeling Chemistry 1st Semester Final V2 3
 - The Rise of Digital Reading Modeling Chemistry 1st Semester Final V2 3
 - Advantages of eBooks Over Traditional Books
2. Identifying Modeling Chemistry 1st Semester Final V2 3
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Modeling Chemistry 1st Semester Final V2 3
 - User-Friendly Interface
4. Exploring eBook Recommendations from Modeling Chemistry 1st Semester Final V2 3
 - Personalized Recommendations
 - Modeling Chemistry 1st Semester Final V2 3 User Reviews and Ratings

- Modeling Chemistry 1st Semester Final V2 3 and Bestseller Lists
- 5. Accessing Modeling Chemistry 1st Semester Final V2 3 Free and Paid eBooks
 - Modeling Chemistry 1st Semester Final V2 3 Public Domain eBooks
 - Modeling Chemistry 1st Semester Final V2 3 eBook Subscription Services
 - Modeling Chemistry 1st Semester Final V2 3 Budget-Friendly Options
- 6. Navigating Modeling Chemistry 1st Semester Final V2 3 eBook Formats
 - ePub, PDF, MOBI, and More
 - Modeling Chemistry 1st Semester Final V2 3 Compatibility with Devices
 - Modeling Chemistry 1st Semester Final V2 3 Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Modeling Chemistry 1st Semester Final V2 3
 - Highlighting and Note-Taking Modeling Chemistry 1st Semester Final V2 3
 - Interactive Elements Modeling Chemistry 1st Semester Final V2 3
- 8. Staying Engaged with Modeling Chemistry 1st Semester Final V2 3
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Modeling Chemistry 1st Semester Final V2 3
- 9. Balancing eBooks and Physical Books Modeling Chemistry 1st Semester Final V2 3
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Modeling Chemistry 1st Semester Final V2 3
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Modeling Chemistry 1st Semester Final V2 3
 - Setting Reading Goals Modeling Chemistry 1st Semester Final V2 3
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Modeling Chemistry 1st Semester Final V2 3
 - Fact-Checking eBook Content of Modeling Chemistry 1st Semester Final V2 3
 - Distinguishing Credible Sources

13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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