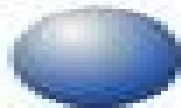


# Matter Practice Assessment

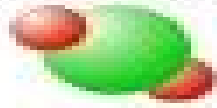
Name: \_\_\_\_\_

**SK000** Label each of the following as an atom, element, or compound

a.



b.



c.



d.



**CS000** Classify the following as homogeneous or heterogeneous mixtures:

a. jar of different colored jelly beans

c. manganese

**SK000** What happens to the properties of elements, when the elements are chemically combined in a molecule?

**SK000** A solution of a mysterious chemical is bright aqua color. If aluminum foil is added to the mysterious chemical solution, the aqua color disappears, a strange odor is released, and the mixture takes on a bright blue color, and gas and a red solid forms.

a. What is an example of a physical property discussed?

b. What is an indication that a chemical change has occurred?

**SK000** Below is a description of a mixture that you are required to separate in to its pure substance components. After considering the "ingredients" in the mixture outline what separation methods you would use, and in what order.

Mix contains: Water, Kool-Aid, glass beads, and benzene (a liquid that does not mix with water).

# Modeling Chemistry Unit Objectives Matter Answers

**Camilla Rothe**



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