Name	
Date	Per

Mixed Practice: Names and Formulas

First determine whether each of the following compounds is *ionic* (M + NM), polyatomic (3 or more atoms), or covalent (2 NM's). Then write the name for each compound.

1) NO ₂		
2) NaBr		
3) SiO ₂		
4) P ₂ Br ₄		
5) FeSO ₄		
6) SF ₆		
7) Li ₂ S		
8) MgBr ₂		
9) N ₂ S		
10) Be(OH) ₂		
11) SO ₃		
12) Cu ₂ S		
13) BF ₃		
14) I ₂		-
15) Pb ₃ (PO ₃) ₂		

Modeling Chemistry Unit 4 Test

Laurence W. McKeen

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