
Maximum Marks: 80

Time allowed Two hours

Answers to this Paper must be written on the paper provided separately

You will not be allowed to write during first 15 minutes.

This time is to be spent in reading the question paper.

The time given at the head of this Paper is the time allowed for writing the answers.

Section A is compulsory. Attempt any four questions from Section B.

The intended marks for questions or parts of questions are given in brackets [].

SECTION A (40 Marks)

(Attempt all questions from this Section.)

Question 1

Choose the correct answers to the questions from the given options.

[15]

(Do not copy the questions, write the correct answers only.)

(i) An element in period 3, whose electron affinity is zero:

- (a) Neon
- (b) Sulphur
- (c) Sodium
- (d) Argon

Answer: (d)

Argon belongs to period 3. It is an inert element and stable. It neither requires gain nor loss. It has its octet complete and hence it has the zero electron affinity.

(ii) An element with the largest atomic radius among the following is:

- (a) Carbon
- (b) Nitrogen

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Table of Contents Model Test Paper For Class Icse 24

1. Understanding the eBook Model Test Paper For Class Icse 24
 - The Rise of Digital Reading Model Test Paper For Class Icse 24
 - Advantages of eBooks Over Traditional Books
2. Identifying Model Test Paper For Class Icse 24
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Model Test Paper For Class Icse 24
 - User-Friendly Interface
4. Exploring eBook Recommendations from Model Test Paper For Class Icse 24
 - Personalized Recommendations
 - Model Test Paper For Class Icse 24 User Reviews and Ratings
 - Model Test Paper For Class Icse 24 and Bestseller Lists
5. Accessing Model Test Paper For Class Icse 24 Free and Paid eBooks
 - Model Test Paper For Class Icse 24 Public Domain eBooks
 - Model Test Paper For Class Icse 24 eBook Subscription Services

- Model Test Paper For Class Icse 24 Budget-Friendly Options
- 6. Navigating Model Test Paper For Class Icse 24 eBook Formats
 - ePub, PDF, MOBI, and More
 - Model Test Paper For Class Icse 24 Compatibility with Devices
 - Model Test Paper For Class Icse 24 Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Model Test Paper For Class Icse 24
 - Highlighting and Note-Taking Model Test Paper For Class Icse 24
 - Interactive Elements Model Test Paper For Class Icse 24
- 8. Staying Engaged with Model Test Paper For Class Icse 24
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Model Test Paper For Class Icse 24
- 9. Balancing eBooks and Physical Books Model Test Paper For Class Icse 24
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Model Test Paper For Class Icse 24
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Model Test Paper For Class Icse 24
 - Setting Reading Goals Model Test Paper For Class Icse 24
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Model Test Paper For Class Icse 24
 - Fact-Checking eBook Content of Model Test Paper For Class Icse 24
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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