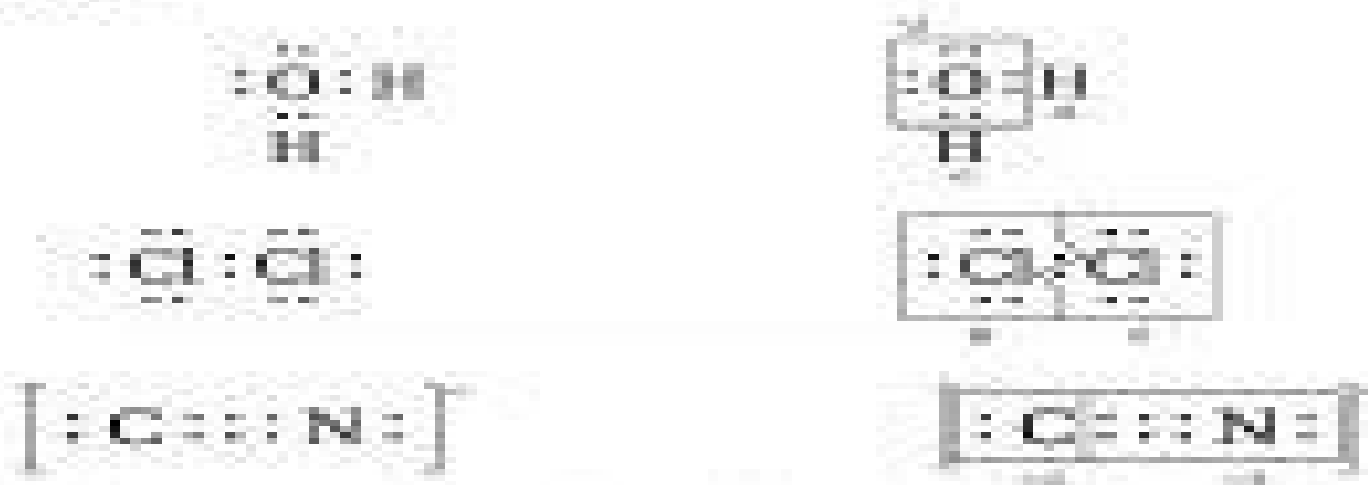


Resonance-Structural Drawing

Resonance Structures:



1. correct electron-distribution, the number of valence electrons, and the overall charge of the molecule, correct formal charges, and

2. correct Lewis arrangement of the atoms.

3. correct formal charges.

There should be a potential for resonance structures.

1. A molecule with resonance structures is any an electron delocalized system, or conjugated. A delocalized system is a molecule that has



2. or

There is one double bond adjacent to

another double bond.

This should be potential for resonance structures.



...resonance structures have equivalent electron density at each carbon. They are electron delocalized with pi bonds.

3. or

There are 4 conjugated double bonds or more.

4. Two adjacent pi bonds.

5. Conjugated pi pi system conjugation system.

1. Conjugated pi pi system system 2. The conjugated system is

3. and pi pi system system 4. The conjugated system is



Mhr Chemistry 12 Aswers

RJ Shavelson



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mechanics as they apply to problems of interest to chemists Stresses the practical computational side of physical chemistry
rather than the theoretical Emphasizing computations it takes the reader step by step through various calculations including
a Hartree Fock calculation of the ground state energy of LiH and a calculation of stationary properties of real gases and
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