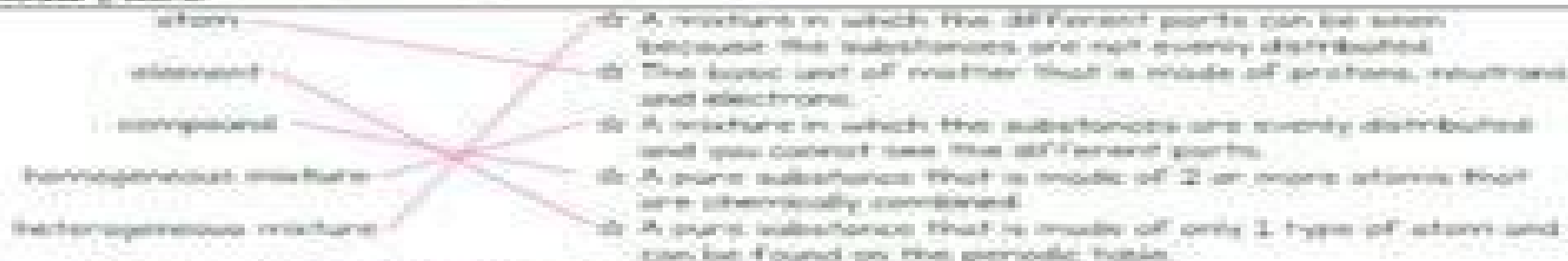


PROPERTIES OF MATTER STUDY GUIDE



ATOMS, ELEMENTS, COMPOUNDS + MIXTURES

Complete the vocabulary matching below by drawing a line to match the correct definition to the correct vocabulary word.



Complete the chart below about the subatomic particles and their charges:

	PARTICLE #1	PARTICLE #2	PARTICLE #3
NAME	proton	neutron	electron
CHARGE	+ (positive)	0 (neutral)	- (negative)
LOCATION	part of nucleus	part of nucleus	outer "cloud-like" orbit

Complete the graphic organizer below about matter below. Need help? Use the word bank to assist you!

heterogeneous	element	mixture	compound
	pure substance	electron	homogeneous



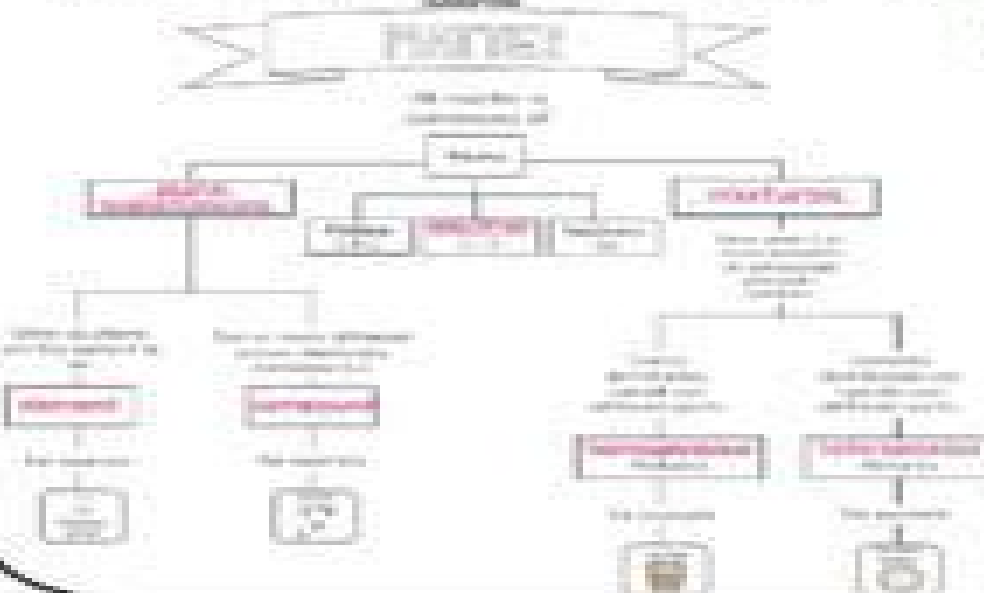
Classify the following examples as elements, compounds or mixtures in the table below.

carbon (C)	salt	water (H ₂ O)
hydrogen (H)	salt water	sugar (sucrose)

ELEMENT	COMPOUND	MIXTURE
carbon (C) hydrogen (H)	water (H ₂ O) salt (NaCl) sugar (sucrose)	salt

Why is carbon considered an element and carbon dioxide considered a compound?

Carbon is an element on the periodic table, made of only carbon atoms.
Carbon dioxide is carbon and oxygen chemically combined together.



Matter Review Guide Answer Key

Y Pai



Matter Review Guide Answer Key:

Unveiling the Magic of Words: A Review of "**Matter Review Guide Answer Key**"

In a global defined by information and interconnectivity, the enchanting power of words has acquired unparalleled significance. Their ability to kindle emotions, provoke contemplation, and ignite transformative change is truly awe-inspiring. Enter the realm of "**Matter Review Guide Answer Key**," a mesmerizing literary masterpiece penned by way of a distinguished author, guiding readers on a profound journey to unravel the secrets and potential hidden within every word. In this critique, we shall delve in to the book is central themes, examine its distinctive writing style, and assess its profound impact on the souls of its readers.

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