Name:	KEY	Block:

Assignment #2 - Compound Names and Formulas Multi-valent ions only

Α_	Write the correct formula for th	e following compounds, all of which have been named using the
		The ionic charge is given after the first element (metallic) in Roman
Nu	merals.	
	C.A	1 0 - 0

1. copper (II) oxide	11. manganese (III) oxide MY 203
2. mercury (I) oxide	12. vanadium (II) bromide VB(z
3. gold (III) chloride AuCl3	13. niobium(V) oxide No205
4. thallium (III) bromide TIPY 3	14. titanium (III) oxide
5. bismuth (V) oxide 81205	15. titanium (III) nitride TIN
6. terbium (III) oxide	16. iron (II) oxide <u>FCO</u>
7. uranium (VI) oxide \(\lambda \)	17. cobalt (II) phosphide C03P2
8. protactinium (IV) bromide PhBY4	18. tin (II) oxide
9. cerium (III) oxide	19. thulium (II) bromide TWBY 2
10. arsenic (V) sulphide A52S5	20. copper (II) bromide CUBY2
B. Name the compound using the Roman Numer	이 그리고 하게 하게 하다가 있다. 그리고 하는 이 이렇게 하게 하게 하는 것이 없는 것이었다면 없어요.
1. SnCl Tin (IV) Chlaride	9. Par. Rolladium (IV) Fluoride
2. BIBro BISMININ (V) BIOMICE	10. Os2O3 OSMIUM (III) Oxide
3. POO2 POLONIUM (IV) OXICE	11. MOBr. MOLYbodenium (II) Bromide
4. Pbllead(II) Iodide	12. VCI6 Vanadium (V) Chloride
5. HgO Mercury (II) Oxide	13. Mn2O3 Managanese (III) Oxide
6. HgCI MEYCUYY (I) ChYOYIDE	14. COO COOOH (II) Oxide
7. Auzos Gold (III) Oxide	15. Np2O3 NEDTUDIUM (II) OXIC
8. FeClo Iran (II) Chloride	16. V2Os Vanadium (V) Oxide

Ionic Compounds Homework Review Sheet Answer Key

Michael Brown

Ionic Compounds Homework Review Sheet Answer Key:

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