

Science 101

# Chapter Test A

## Teacher Notes and Answers

### Subatomic Physics

#### Questions 1-10 are (Answers)

1. A
  2. A
  3. A
  4. B
  5. B
  6. A
  7. B
  8. A
  9. A
  10. A
- Answers:
1. Mass number of a nucleus is the sum of the number of protons and neutrons in the nucleus.
2. The mass number of a nucleus is the sum of the number of protons and neutrons in the nucleus.
3. The mass number of a nucleus is the sum of the number of protons and neutrons in the nucleus.
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11. The mass number of a nucleus is the sum of the number of protons and neutrons in the nucleus.
12. The mass number of a nucleus is the sum of the number of protons and neutrons in the nucleus.
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14. The mass number of a nucleus is the sum of the number of protons and neutrons in the nucleus.
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16. The mass number of a nucleus is the sum of the number of protons and neutrons in the nucleus.
17. The mass number of a nucleus is the sum of the number of protons and neutrons in the nucleus.
18. The mass number of a nucleus is the sum of the number of protons and neutrons in the nucleus.
19. The mass number of a nucleus is the sum of the number of protons and neutrons in the nucleus.
20. The mass number of a nucleus is the sum of the number of protons and neutrons in the nucleus.

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