

Physical Science Standards:

SPS1: Students will investigate our current understanding of the atom.

a. Examine the structure of the atom in terms of

- proton, electron, and neutron locations.
 - Proton has a charge of _____ and is located in the _____ of the atom.
 - Protons are also the elements' _____ number of the identity of the element, so protons are found by looking at the atomic number
 - Neutron has a charge of _____ and is located in the _____ of the atom.
 - Neutrons are found by subtracting the _____ Number from the _____ Mass or the BIG #- SMALL #.
 - Electron has a charge of _____ and is located in the _____ of the atom.
 - Electrons are equal to the number of _____ in a neutral atom.
 - The nucleus is made up of these 2 particles, _____ and _____.
 - The electrons are located in this region of the atom, known as the _____.
 - The electron cloud is divided into energy levels or energy shells, the first level will hold _____ electrons and the second level will hold _____ electrons.
 - The valence shell or valence level is the _____ energy shell of the atom.
- atomic mass and atomic number.
 - Atomic number is equal to the number of _____ in the atom and is also the atom's identity—it's the small number of the periodic table.
 - Atomic mass is equal to the average mass of all the isotopes of a particular element—it's the big number on the periodic table.
- atoms with different numbers of neutrons (isotopes).
 - Define Isotope _____.
 - Ex. Carbon-12 and Carbon-14
 - Carbon 12 has _____ P _____ N _____ E
 - Carbon 14 has _____ P _____ N _____ E
- explain the relationship of the proton number to the element's identity.
 - The _____ number is equal to the number of _____ in the atom which is how the element is identified.
 - Ex. Nitrogen has an atom number of _____, which means that it has _____ protons.

molybdenum	← element name
42	← atomic number number of protons (Z)
Mo	← atomic symbol
95.94	← atomic mass A (this is an average mass)



- Number of Protons = Atomic Number
- Number of Electrons = Number of Protons = Atomic Number
- Number of Neutrons = Mass Number - Atomic Number

Eoct Physics Study Guide

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Eoct Physics Study Guide:

This book delves into Eoct Physics Study Guide. Eoct Physics Study Guide is a vital topic that must be grasped by everyone, ranging from students and scholars to the general public. The book will furnish comprehensive and in-depth insights into Eoct Physics Study Guide, encompassing both the fundamentals and more intricate discussions.

1. This book is structured into several chapters, namely:

- Chapter 1: Introduction to Eoct Physics Study Guide
- Chapter 2: Essential Elements of Eoct Physics Study Guide
- Chapter 3: Eoct Physics Study Guide in Everyday Life
- Chapter 4: Eoct Physics Study Guide in Specific Contexts
- Chapter 5: Conclusion

2. In chapter 1, the author will provide an overview of Eoct Physics Study Guide. The first chapter will explore what Eoct Physics Study Guide is, why Eoct Physics Study Guide is vital, and how to effectively learn about Eoct Physics Study Guide.
3. In chapter 2, the author will delve into the foundational concepts of Eoct Physics Study Guide. The second chapter will elucidate the essential principles that need to be understood to grasp Eoct Physics Study Guide in its entirety.
4. In chapter 3, the author will examine the practical applications of Eoct Physics Study Guide in daily life. The third chapter will showcase real-world examples of how Eoct Physics Study Guide can be effectively utilized in everyday scenarios.
5. In chapter 4, this book will scrutinize the relevance of Eoct Physics Study Guide in specific contexts. The fourth chapter will explore how Eoct Physics Study Guide is applied in specialized fields, such as education, business, and technology.
6. In chapter 5, this book will draw a conclusion about Eoct Physics Study Guide. This chapter will summarize the key points that have been discussed throughout the book.

This book is crafted in an easy-to-understand language and is complemented by engaging illustrations. This book is highly recommended for anyone seeking to gain a comprehensive understanding of Eoct Physics Study Guide.

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