

## Stoichiometry Worksheet 2: Percent Yield

What aspects of the procedure do you think:

- affected the theoretical or actual yield negatively?
- substantially affected your yield in a positive or negative way? explain how these things can be made up.
- affected your setup or the analysis. (Cleaning up flasks, glass break, incomplete transfer, incorrect mass balance)

10. Using the theoretical mass of the silver chloride, calculate the theoretical mass of silver chloride. (What is the percent yield?) (What is the percent yield?) (What is the percent yield?) (What is the percent yield?)

$$\text{Mass of Silver Chloride} = \frac{\text{Mass of Silver Chloride}}{\text{Mass of Silver Chloride}} \times \text{Mass of Silver Chloride}$$

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11. Calculate the percent yield of silver chloride. (What is the percent yield?) (What is the percent yield?) (What is the percent yield?) (What is the percent yield?)

$$\text{Percent Yield} = \frac{\text{Mass of Silver Chloride}}{\text{Mass of Silver Chloride}} \times 100\%$$

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12. What mass of silver chloride is formed? (What is the percent yield?) (What is the percent yield?) (What is the percent yield?) (What is the percent yield?)

$$\text{Mass of Silver Chloride} = \frac{\text{Mass of Silver Chloride}}{\text{Mass of Silver Chloride}} \times \text{Mass of Silver Chloride}$$

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13. Calculate the percent yield of silver chloride. (What is the percent yield?) (What is the percent yield?) (What is the percent yield?) (What is the percent yield?)

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**M Walker**



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