

Stoichiometry Worksheet 2: Percent Yield

What aspects of a the percent yield calculation:

- Identify a balanced chemical equation of a reaction.
- Calculate the actual yield of a reaction (actual yield) from a given amount of reactant (theoretical yield).
- Identify the theoretical yield of a reaction (theoretical yield) from a given amount of reactant.

1. Calculate the theoretical yield of a reaction (theoretical yield) from a given amount of reactant (theoretical yield).
 Identify the reactant (theoretical yield) from a given amount of reactant (theoretical yield).

$$\text{Reactant} \rightarrow \text{Product} \quad \text{or} \quad \text{Reactant} \rightarrow \text{Product} \quad \text{or} \quad \text{Reactant} \rightarrow \text{Product}$$

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2. Calculate the percent yield of a reaction (percent yield) from a given amount of reactant (theoretical yield) and a given amount of product (actual yield).
 Identify the reactant (theoretical yield) from a given amount of reactant (theoretical yield).

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4. Calculate the percent yield of a reaction (percent yield) from a given amount of reactant (theoretical yield) and a given amount of product (actual yield).
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Modeling Chemistry U8 Ws 1 V2 0

Jessica J Manson



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