Mixed Mole Problems Worksheet

Directions: Solve the following problems. Be sure to show all your work

- 1. Find the number of moles if there are 7.63×10^{24} molecules of a substance.
- 2. How many moles of CaCO3 are there in 2.00 kg of CaCO3?
- 3. What is the mass of 8.42 moles of (NH4)≥5?
- 4. Find the number of carbon atoms in 18.5 q of CH3OH (methanol).
- 5. If one litre of a gas at STP has a mass of 3.17 g, what is the molar mass of this gas?
 Name the diatomic element that makes up this gas?

- 6. What volume at STP would 11.50 g of oxygen gas occupy?
- 7. What is the mass of 33.2 L of SO2 gas at STP?
- 8. Calculate the molar mass of a gas at STP, if 360.0 mL of this gas has a mass of 0.680 g

Mixed Mole Problems Show Work

Kathleen E. Murphy

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