Quiz)

Name ANSWER WEYERS

Date

Co HIDO

Assessment

Ions in Aqueous Solutions and Colligative Properties

Section Quiz: Colligative Properties of Solutions

In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question.

- 1. Why is freezing-point depression a colligative property?
 - a. It is inversely proportional to the number of particles in a solution.
 - b. It is directly proportional to the number of particles in a solution.
 - c. It depends on the properties of an electrolyte in a solvent.
 - d. None of the above
- Compared with a 0.01 M C₀H₁₂O₀ solution, a 0.01 M KCl solution has
 - a. the same freezing-point depression.
 - b. about twice the freezing-point depression.
 - c. the same freezing-point elevation.
 - d. about six times the freezing-point elevation.
- 3. Compared with a 1.00 M NaI solution, a 1.00 M Na₂SO₄ solution has
 - a. the same boiling-point elevation.
 - b. about twice the boiling-point elevation.
 - c. a boiling-point elevation about two-thirds as high.
 - d. a boiling-point elevation about 1.5 times as high.
- 4. When a nonvolatile solute dissolves in a solvent, the vapor pressure of the solvent
 - a. increases.
 - b. decreases.
 - c. stays the same.
 - changes depending on the solvent used.
- 5. Which of the following compounds would be most effective in lowering the melting point of ice on roads?
 - a. CaCl.,
 - b. NaCl.
 - c. K₀PO₄
 - d. K₂SO₄

Nacl Hio Nat

(9) + C/F)

VAPOR PRESSURE LOWERING (Colligative

PO 7/2 (

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Modern Chemistry

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Quite

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Robert J. Sager

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Math Ch 13 Quiz 2 Answer Key Book Review: Unveiling the Magic of Language

In an electronic digital era where connections and knowledge reign supreme, the enchanting power of language has be apparent than ever. Its power to stir emotions, provoke thought, and instigate transformation is truly remarkable. This extraordinary book, aptly titled "Math Ch 13 Quiz 2 Answer Key," published by a highly acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound impact on our existence. Throughout this critique, we will delve in to the book is central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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