

Quiz & Worksheet - Modern Atomic Theory & Theorists

1. Near the end of the 18th century the first atomic theory was developed. Which of the following was not part of that early understanding?

- ☐ All matter is made up of atoms.
- ☐ Elements are composed of only one type of atom.
- ☐ Atoms can combine with other atoms to make more complex substances.
- ☐ Electrons orbit the nucleus in electron clouds.
- ☐ All of these concepts were understood in the 18th century.

2. Heisenberg proposed the idea that if an electron moves as a wave, it would be impossible to simultaneously measure both its position and momentum. This idea is known as which of the following?

- ☐ The Modern Atomic Theory
- ☐ The First Atomic Theory
- ☐ The Uncertainty Principle
- ☐ The Certainty Principle
- ☐ The Heisenberg Model

3. Which model can be described as negatively-charged electrons orbiting a positively-charged nucleus in definite paths?

- ☐ The Bohr Model
- ☐ The Schrodinger Model
- ☐ The Schrodinger Equation
- ☐ The Heisenberg Model
- ☐ The Heisenberg Principle

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