

Chemistry Unit 8 Worksheet 4

Samples of Every Kind of Problem

Let us suppose that our "paper" writer is a chemist who has to write out what products may be made. What are the possibilities and how do you find them? The next 4 problems cover typical questions.

1. Chlorine reacts with bromine in a reaction of gaseous molecules. $\text{Br}_2 + \text{Cl}_2$ say that some decomposition for products produces chlorine, Br_2 say, and 1.0 moles of oxygen gas.

Reactants	1.0 Br_2	1.0 Cl_2	0	0
Products	0	0	1.0	0
Change	1.0	1.0	-1.0	0

1.0 moles of Br_2 and 1.0 moles of Cl_2

2. How much a displacement reaction, magnesium metal reacts with hydrochloric acid to produce magnesium chloride and hydrogen gas. How much of hydrochloric acid is required to produce 1.0 mole of H_2 gas?

Reactants	0	1.0 H_2	0	0
Products	0	0	1.0	0
Change	0	1.0	-1.0	0

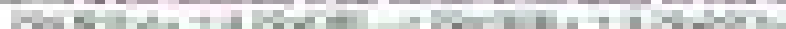
1.0 moles of H_2 gas

3. Chlorine, Cl_2 , reacts with hydrogen gas to produce hydrogen chloride gas. How much Cl_2 is required to produce 1.0 moles of hydrogen gas by displacement of a metal with 1.0 moles of metal?

Reactants	0	1.0 Cl_2	0	0
Products	0	0	1.0	0
Change	0	1.0	-1.0	0

1.0 moles of Cl_2 gas

4. Chlorine reacts with hydrogen gas to produce hydrogen chloride gas. How much Cl_2 is required to produce 1.0 moles of hydrogen gas by displacement of a metal with 1.0 moles of metal?



1.0 moles of Cl_2 gas

Reactants	0	1.0 Cl_2	0	0
Products	0	0	1.0	0
Change	0	1.0	-1.0	0

1.0 moles of Cl_2 gas

Modeling Chemistry Unit 8 Formula Practice

Puja Mehta

A decorative graphic consisting of a light blue horizontal bar with a rounded right end, and a red circular shape partially visible behind it.

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