#### CHAPTER 7 PRACTICE TEST: ATOMIC STRUCTURE & PERIODIC TRENDS- KEY

1)When a solution of	sodium chloride	is vaporized in a flame, the	he color of the flame is:	
A) Blue B) Yel	low C) Green	D) Violet E) V	White	
2) Electromagnetic ra	diation travels th	rough vacuum at a speed	of m/s.	
A) 186,000	C) 3.00×10 <sup>8</sup>	E) It depends or	n .	
B) 125	D) 10,000	λ		
3) The wavelength of	light that has a fi	requency of 1.20×10 <sup>(3</sup> s <sup>-1</sup>	ism.	
A) 25.0	B) 2.50×10 <sup>-5</sup>	C) 0.0400 D) 12.0	E) 2.5	
GIVEN: $\nu = 1.20 \times 10$	0 <sup>13</sup> s <sup>-1</sup> U	INKNOWN: A		
CONNECTION: e = /	£ 14.5			
SUBSTITUTE & SOLVE	$\lambda = c/\nu$			
$\lambda = 3.00 \times 10^8 \text{ m} \bullet \text{ s}^{-1}$ $1.20 \times 10^{13} \text{ s}^{-1}$	i.			
$\lambda = 2.50 \times 10^{-5} \mathrm{m}$				
A) 500 GIVEN: λ = 6 m CONNECTION: c = λ SUBSTITUTE & SOLVE ν = 3.00 x 10 <sup>8</sup> m • s	B) 200 UNKNO $\ell \nu = \epsilon/\lambda$	WN: v (Must convert and	he frequency of this EM ra D) 20 swer to MHzI)	ediation is MHz. E) 2.0
i = 0.5 x 10 <sup>8</sup> Hz →	10° Hz	50 MHz		
5) The energy of a phoits wavelength.	oton of light is _	proportional t	o its frequency and	proportional to
A) directly, directly		C) inversely, directly	E) indirect	tly, not
B) inversely, inversely	Y	D) directly, inversely		
		avelength of 12.3 nm is _	J.	
A) 1.51×10 <sup>-17</sup>	B) 4.42×10 <sup>-23</sup>	C) 1.99×10 <sup>-24</sup>	D) 2.72×10 <sup>-50</sup>	E) 1.62×10 <sup>-17</sup>
	a photon that has	an energy of 5.25×10-19	J is m.	
A) 3.79×10 <sup>-7</sup>	B) 2.64×10 <sup>6</sup>	C) 2.38×10 <sup>23</sup>	D) 4.21×10 <sup>-24</sup>	E) 3.79×10 <sup>™</sup>
8) A mole of red photons of wavelength 725 nm has			kJ of energy.	
A) 2.74×10 <sup>-19</sup>	B) 4.56×10 <sup>-66</sup>	C) 6.05×10 <sup>-3</sup>	D) 165	E) 227
$\lambda = 725  \text{nm}  \text{ x}$		m before using formulas 10 <sup>-7</sup> m	)	

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**G** Thomas

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