Level - I (Based on NCERT)

 Which is not an example of exothermic reaction: (a) N₂ + O₂ → 2NO - Heat (b) N₂ + 3H₂ →2NH₃ + Heat (c) H₂ + I₃ →2HI + Heat (d) All of these 2. Which is an example of endothermic reaction: (a) N₂ + O₂ →2NO - Heat (b) CaCO₃ → CaO + CO₂ − Heat (c) 2KClO₃ → 2kCl + 3O₂ − Heat (d) All of these Which is an example of combination reaction: (a) N₂ + Cl₂ → 2HCl (b) C + O₂ → CO₂ (d) All of these (c) $N_2 + 3 H_2 \rightarrow 2NH_3$ 4. Which is an example of decomposition reaction? (a) 2Pb(NO₃)₂ → 2PbO + 4NO₂ + O₂ (b) 2FeSO₄ → Fe₂O₄ + SO₂ + SO₃ (c) 2H2O Electricity 2H2+ O2 (d) All of these 5. Which is an example of thermal decomposition? (b) 2Al₂O₃ Electricity 4Al + 3O₃ (a) 2NaCl Electricity 2Na + Cla (c) CaCO₃ → CaO + CO₂ (d) NaOH + HCl →NaCl + H₂O 6. Which is an example of electrolysis decomposition? (a) 2KClO₃ → 2KCl + 3O₂ (b) BaCl₂ + Na₂SO₄ → BaSO₄ + 2NaCl (d) 2Al₂O₂(I) Electricity 4Al + 3O. (c) 2AgCl hv2Ag + Cl2 7. Which is an example of photo chemical reactions? (a) H₂ + C12 Ker 2 HC1 (b) 2AgBr W 2Ag + Br2 (c) 6CO2 + 6H2O W C6H12O6 (d) All of these 8. Which compound is used in photography? (b) AgCI (c) Na₂CO₂ (d) (a) and (b) 9. Which is an example of double displacement reaction? (a) BaCl₂ + Na₂SO₄ → BaSO₄ + 2NaCl (b) AgNO₃ + NaCl →AgCl + NaNO₃ (c) NaOH + HCl →NaCl + H₂O (d) All of these 10. Fe₂O₃ + 2Al →Al₂O₃ + 2Fe is an example of (a) Combination reaction (b) Decomposition reaction (c) Double displacement reaction (d) Displacement reaction 11. Which is an example of double displacement reaction? (a) Mg + 2HCl → MgCl₂ + H₂ (b) ZnCO₃ → ZnO + CO₂ (c) H₂ + Cl₂ →2HCl (d) BaCl₂ + K₂SO₄ → BaSO₄2KCl 12. H₂S + Cl₂ →2HCl + S is this reaction 'Cl₂' act as: (a) Oxidising agent (b) Reducing agent (d) None of these (c) Both (a) & (b) Which statement is correct regarding the reaction 2PbO + C → C₂Pb + CO₂ (a) Lead is getting reduced (b) Carbon dioxide is getting oxidised (c) Lead oxide is getting reduced (d) All of these 14. In the reaction MnO₂ + 4HCl → MnCl₂ + 2H2O + Cl₂, MnO₂ act as? (a) Reducing agent (b) Oxidising agent (c) Both (a) & (b) (d) None of these 15. Which is an example of redox reaction? (a) Zn + CuSO₄ → ZnSO₄ + Cu (b) H₂S + I2 → 2HCl + S (c) CuO + H₂ → Cu + H₂O (d) All of these 16. Which is a formula of rust? (a) Fe₂O₃ (b) Fe₂O (c) Fe₂O₃×H₂O (d) FeCO₁ 17. Rusting of Iron is an example: (a) Oxidation (b) Reduction (c) Redox (d) All of these 18. Antioxidant is/are: (d) None of these (a) Oxidising agent (b) Reducing agent (c) Both (a) & (b) 19. Rancidity can be prevented by: (a) Adding antioxidants (b) Storing food in air light

Modeling Chemestry Unit 10 Ws 2 Answers

L Reisser

Modeling Chemestry Unit 10 Ws 2 Answers:

Modeling Chemestry Unit 10 Ws 2 Answers Book Review: Unveiling the Power of Words

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