

Name: \_\_\_\_\_

Quiz Date: \_\_\_\_\_

### **Mixtures & Solutions I Check Investigation 1**

(p. 3-14 in science resources book and lab notes from science notebook)

Be able to argue the difference between a mixture and a solution

How could you separate a mixture like you separated the salt, gravel, and powder in class?

Could salt and water be both a mixture and a solution?

When a solid material dissolves in a liquid to form a solution, the solid material is called the \_\_\_\_\_.

Be able to tell how a dry mixture could be separated. (like the lab we did in class)

Would a sugar and water solution be able to be separated? How? Why?

### **Mixtures & Solutions Investigation 2 I Check Study Guide**

Quiz Date: \_\_\_\_\_

Know the vocabulary

Review the content questions and answers

Reread *Solutions Up Close*, *The Bends*, and *Summary: Reaching Saturation*

Know how crystals form

How could you figure out how much sugar could saturate water?

If you lost count of spoonfuls, how else could you measure how much sugar could saturate the water solution?

What evidence would you look for to find out that salt and water make a solution?

How do you know when a solution is saturated?

Compare and contrast Sodium Chloride and Epsom Salts as they dissolve in water

### **Mixtures & Solutions I Check 3 Study Guide**

Know and understand the vocabulary words for Investigation 3, reread the science textbook (pages 28-40), review your science lab book, and watch the teacher prep videos for this investigation at:

<http://www.fossweb.com/modules3-6/MixturesandSolutions/index.html>

Watch Investigation 4 not 3 (they are switched on the website)

What is a chemical reaction?

# Mixture And Solution Study Guide

**Mark Steyvers**



## Mixture And Solution Study Guide:

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