

Gay-Lussac's Law

Ex. (3) What would be the new temperature of a rigid container of a gas at 101.3 kPa and 293 K if the gas pressure increased to 115 kPa?

$$\frac{P_1}{T_1} = \frac{P_2}{T_2}$$

$$\frac{(101.3 \text{ kPa})}{(293 \text{ K})} = \frac{(115 \text{ kPa})}{(X)}$$

[X = 333 K]

Gay Lussac S Worksheet Answers

Wyatt William Randall

Gay Lussac S Worksheet Answers:

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Exploration of the Structure of Atom Chandan Sengupta, First Publication April 2019 Revised Publication October 2022 Second Revised Edition July 2023 Third Revised Edition December 2024 Published by Chandan Sengupta Printed by IECIT Printing and Publication Services Resource Centre Arabinda Nagar Bankura 722101 W B India Topics Covered Atoms and Molecules Structure of Atom Bonding Mechanism and Chemical Reactions Mechanism of Bonding This book is suitable for students of Class 9 to Class 11 Students aspiring for Pre Medical Entrance Examination can also get adequate support Additional Hard Copies can also be obtained from Chandan Sukumar Sengupta Arabinda Nagar Bankura 722101 WB Write to Us for more materials

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