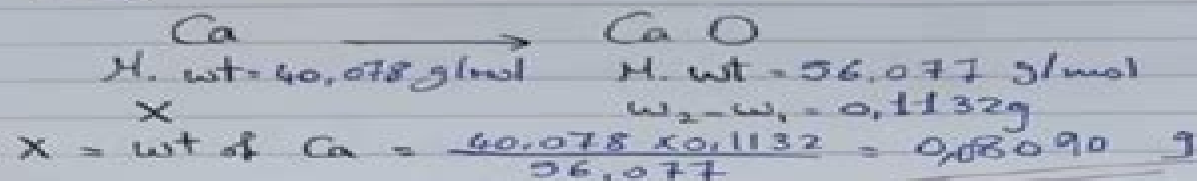


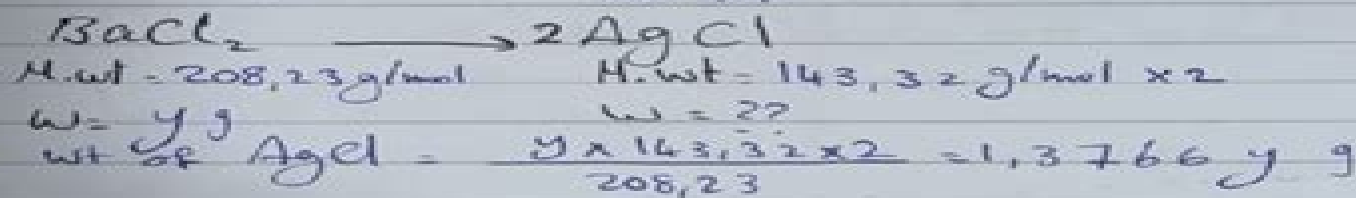
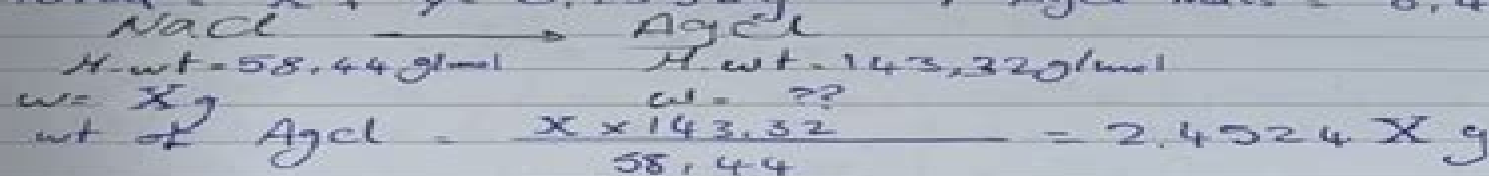
## problems (1)

- 1)  $w_1 = 26.6002 \text{ g}$  ,  $w_1 = \text{mass of empty crucible}$   
 $w_2 = 26.7134 \text{ g}$  ,  $w_2 = \text{mass of crucible and CaO}$   
 $w_{\text{of CaO}} = w_2 - w_1 = 26.7134 - 26.6002 = 0.1132 \text{ g}$



$X \longrightarrow 200 \text{ mL}$   
 $\text{Strength} \longrightarrow 100 \text{ mL}$   
 $\therefore \text{Strength of Ca} = \frac{0.08090 \times 100}{200} = 0.04045 \text{ g/100 mL sample}$

- 2)  $X = \text{mass of NaCl}$  ,  $y = \text{mass of BaCl}_2$   
 $\text{total} \cdot X + y = 0.2356 \text{ g}$  ,  $\text{AgCl mass} = 0.4637 \text{ g}$



$\therefore \text{wt of AgCl} = 2.4524 X + 1.3766 y$   
 $\phantom{\therefore \text{wt of AgCl}} = 2.4524 X + 1.3766 (0.2356 - X) = 0.4637 \text{ g}$   
 $\therefore X = \text{mass of NaCl} = 0.1296 \text{ g} \quad \therefore \text{NaCl \%} = 55.01 \%$   
 $\therefore y = \text{mass of BaCl}_2 = 0.106 \text{ g} \quad \therefore \text{BaCl}_2 \% = 44.99 \%$

# Gravimetric Analysis Problems With Answers

**J Ma**



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