

**Exercise 1.1 (Thermodynamic potentials)**

The Helmholtz free energy  $A$  of a system is defined as  $A = U - TS$ , where  $U$  is the internal energy,  $T$  is the temperature, and  $S$  is the entropy. The Helmholtz free energy is a function of temperature  $T$  and volume  $V$ . The Helmholtz free energy is a thermodynamic potential. The Helmholtz free energy is a function of temperature  $T$  and volume  $V$ . The Helmholtz free energy is a thermodynamic potential. The Helmholtz free energy is a function of temperature  $T$  and volume  $V$ . The Helmholtz free energy is a thermodynamic potential.

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$$A(T, V) = U(T, V) - TS(T, V)$$

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