Gravimetric Analysis of a Chloride Salt (G)

Date submitted:10/12/2021

By: Ilyaas Hussein (101230436)

Lab section: LW3

TA: Kailai Wang

Purpose:

The purpose of the lab is to find the amount of chloride present in any unknown soluble salt, by showing and explaining the techniques present in gravimetric analysis.

Theory:

 This is the equation that describes the reaction between silver ion and chloride ion to form silver chloride:

$$Ag^{+}(aq) + C\Gamma(aq) \rightarrow AgCl(s)$$

ii. AgCl solubility in water and the Ksp.

"Silver chloride is very insoluble, so we can assume that the above reaction goes to completion." 1

$$K_{SD} = 1.6 \times 10^{-10}$$

- AgCl is precipitated in an acidic medium to reduce the chance of acid ions interfering with the results. These ions will form co-precipitates with silver in acid. Due to these coprecipitates, the results would be higher.
- iv. In order to ensure the precipitate does not go through the filter paper, which would give us lower results. To prevent this from happening, it is heated to make coagulate it into a clumpy mass. Heating the precipitate makes it coagulate, making it much harder for any precipitate to go through the filter paper.
- Silver chloride decomposes into silver and chloride in the presence of light, which is described by the following equation:

$$AgCl_{(s)} \rightarrow Ag_{(s)} + \frac{1}{2}Cl_{(g)}$$

If this photodecomposition takes place in air, it will result in lower results since chlorine will be released as a gas. If it takes place in the presence of excess silver ion in aqueous solution, it will result in higher results, due to chlorine reacting with the silver to form silver chloride and solid silver from the decomposition remaining in the precipitate, which will make the results high. The following chemical equation is a result of chlorine reacting with silver to form silver chloride:

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Sebastian Brünink

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English in Analytical Chemistry. Communicating about Methods & Techniques. Книга для студента Надежда Зорина, Александра Соболева, 2022-10-19 Laboratory Experiments for Brown and LeMay, Chemistry, the Central Chemical Analyses of Plutonium and Plutonium Alloys Rose **Science** John Henry Nelson, Kenneth C. Kemp, 1977 Investigation in General Chemistry Alice S. Cohen, 1978 Radioactive Waste Management ,1995 Working with Chemistry Donald J. Wink, Sharon Fetzer-Gislason, Julie Ellefson Energy Research Abstracts ,1990 Kuehn, 2004-02-20 With this modular laboratory program students build skills using important chemical concepts and techniques to the point where they are able to design a solution to a scenario drawn from a professional environment The scenarios are drawn from the lives of people who work with chemistry every day ranging from field ecologists to chemical engineers and include many health professionals as well Nuclear Science Abstracts ,1974 U.S. Government Research Reports ,1961 Bulletin .1977 U.S. Government Research and Development Reports ,1965 **Selected Water Resources Abstracts** ,1983 INIS Atomindex ,1988 Current Chemical Papers, 1956 A classified world list of new papers in pure chemistry Government Reports Announcements & Index ,1986 **Air Pollution** Abstracts ,1970 Advances in Portable X-ray Fluorescence Spectrometry B Lee Drake, Brandi L MacDonald, 2022-10-19 Over the last two decades advances in the design miniaturization and analytical capabilities of portable X ray fluorescence pXRF instrumentation have led to its rapid and widespread adoption in a remarkably diverse range of applications in research and industrial fields The impetus for this volume was that as pXRF continues to grow into mainstream use analysts should be increasingly empowered with the right information to safely and effectively employ pXRF as part of their analytical toolkit This volume provides introductory and advanced level users alike with readings on topics ranging from basic principles of pXRF and qualitative and quantitative approaches through to machine learning and artificial intelligence for enhanced applications It also includes fundamental guidance on calibrations the mathematics of calculating uncertainties and an extensive reference index of all elements and their interactions with X rays Contributing authors have provided a wealth of information and case studies in industry specific chapters These sections delve into detail on current standard practices in industry and research including examples from agricultural and geo exploration sectors research in art and archaeology and metals industrial and regulatory applications As pXRF continues to grow in use in industrial and academic settings it is essential that practitioners continue to learn share and implement informed and effective use of this technique This volume serves as an accessible guidebook and go to reference manual for new and experienced users in pXRF **Government Reports Announcements** ,1968 Ores and Metals ,1906 to achieve this goal **TID.** ,1953

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