



Nuclear Equations

(a) Disintegration of radium-236 by emission of an alpha particle (${}^4_2\text{He}$).



<i>Mass Nos :</i>	Reactants = 236 ;	Products = 232 + 4 = 236
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<i>Atomic Nos :</i>	Reactants = 88 ;	Products = 86 + 2 = 88
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(b) Disintegration of phosphorus-32 by emission of a beta particle (${}^0_{-1}e$).



<i>Mass Nos :</i>	Reactants = 32 ;	Products : 0 + 32 = 32
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<i>Atomic Nos :</i>	Reactants = 15 ;	Products : 16 - 1 = 15
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(c) Fission of argon-40 by bombardment with a proton (${}^1_1\text{H}$).



<i>Mass Nos :</i>	Reactants = 40 + 1 = 41 ;	Products = 40 + 1 = 41
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<i>Atomic Nos :</i>	Reactants = 18 + 1 = 19 ;	Products = 19 + 0 = 19
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Mcq Nuclear Equation

Michael Brown



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