

## Types of Nuclear Radiation – Questions

**Instructions:** Identify the following descriptions as alpha, beta, or gamma:

- |                                                                            |              |
|----------------------------------------------------------------------------|--------------|
| 1) This radiation can only be stopped by <u>thick concrete or lead</u> .   | <u>Gamma</u> |
| 2) This particle can be stopped by <u>clothing alone</u> .                 | <u>Alpha</u> |
| 3) During this decay, <u>no protons, neutrons, or electrons are lost</u> . | <u>Gamma</u> |
| 4) During this type of decay, <u>two protons</u> are lost.                 | <u>Alpha</u> |
| 5) This particle has a <u>-1 charge</u> .                                  | <u>Beta</u>  |
| 6) This particle has a mass of 4.                                          | <u>Alpha</u> |
| 7) This type of radiation is identical to an <u>electron</u> .             | <u>Beta</u>  |
| 8) This form of radiation has <u>no mass</u> and <u>no charge</u> .        | <u>Gamma</u> |
| 9) This particle is represented as a <u>helium</u> atom.                   | <u>Alpha</u> |
| 10) This particle is emitted as in the form of waves.                      | <u>Gamma</u> |
| 11) This particle can be stopped by a <u>piece of paper</u> .              | <u>Alpha</u> |
| 12) This particle has average (medium) penetrating power.                  | <u>Beta</u>  |
| 13) This is the <u>least dangerous</u> type of radiation.                  | <u>Alpha</u> |
| 14) This is the <u>most dangerous</u> type of radiation.                   | <u>Gamma</u> |

**Instructions:** Use the nuclear equation below to answer the following questions:



- |                                                                     |                    |
|---------------------------------------------------------------------|--------------------|
| 15) The picture above represents <u>what type</u> of nuclear decay? | <u>Beta Decay</u>  |
| 16) What is the mass of the starting phosphorus (P) atom?           | <u>32</u>          |
| 17) What is the mass of the particle released above?                | <u>0</u>           |
| 18) Which decay would have caused a larger change in mass?          | <u>Alpha Decay</u> |

# Key Terms Radioactivity And Nuclear Reactions

## Worksheet Answer

**Howard Stephen Stoker**



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