

Marking Scheme PP1

Q1 (a) Because the almost colourless region of a non-luminous flame contains the unburnt gas which will continue burning at the end of the glass tube when ignited. (2mk)

- (b) - Non-luminous flame produces much heat / it is the hottest ✓
- The flame does not produce soot / smoke which makes glass apparatus dirty. (1mk)

Q2 (a) An acid is a proton donor ✓ (1mk)

(ii) A Base is a proton acceptor ✓ (1mk)

(b) H_2O is the acid ✓ (1mk)

Q3 (a) Graham's law of diffusion state the rate of diffusion of a gas is inversely proportional to the square root of its density provide the temperature and pressure are constant. ✓ (1mk)

(b) $50cm^3$ of $O_2(g) \longrightarrow 80 \text{ secs}$
 $150cm^3$ of $O_2(g) \longrightarrow \frac{1}{2}$

$$\left(\frac{150}{50} \times 80\right) = 240 \text{ secs } \frac{1}{2}$$

$$\frac{r_{O_2}}{r_{SO_2}} = \sqrt{\frac{RMM_{O_2}}{RMM_{SO_2}}}$$

$$\left(\frac{240}{150}\right)^2 = \left(\sqrt{\frac{32}{64}}\right)^2 \frac{1}{2}$$

$$\frac{240 \times 240}{r_{SO_2}^2} = \frac{32}{64} \frac{1}{2}$$

$$r_{SO_2}^2 = 115200 \frac{1}{2}$$

$$r_{SO_2} = 339.41 \frac{1}{2}$$

(3mk)

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M Mark



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