Summative Test in Grade 10 Science Name: Score Name of Teacher Section: Instruction: Write the letter of the correct answer before the number. 1. Chemical reactions C. create and destroy atoms A. Occur only in living things B. Produce new substances D. only occur outside living organisms What are the coefficients that will balance the skeleton equation? AlCl3 + NaOH → Al(OH)3 + NaCil B. 3, 1, 3, 1 C. 1, 1, 1, 3 D. 1, 3, 3, 1 3 In every balanced chemical equation, each side of the equation has the same number of C moles D coefficients A. Atoms in each element B. molecules 4. Which molecules are most closely related to lipids? A. nucleotides B. fatty acids C. amino acida D. sugars 5. Which of the following molecules are classified as carbohydrates? A. mucleotides B. fatty acids C. amino acida D. sugars 6. Polysaccharides are _ A. limids B. carbohydrates D. mucleic acids C. proteins 7. Which molecules are most closely related to proteins? A nucleotides B. fatty acids C. smano acida D. sugars 8. Which of the following is not an example of matter? B. heat C smodee D. water vapor 9. When the equation Fe + Cl2 -> FeCl3 is balanced, what is the coefficient for Cl2 ? A. 1 C: 3 D 4 10. What are the missing coefficients for the skeleton equation below? Al2 (SO4)3 (aq) KOH(aq) → Al(OH)3 (aq) K2 SO4 (aq) B. 2,12,4,6 C 4.6.2.3 A. 1.3.2.3 11 The equation above is an example of what type of chemical reaction? A. combination B. decomposition C. single displacement D. double displacement. 12. The type of reaction that takes place when one element reacts with a compound to form a new compound and a different element is a reaction. B. decomposition A. combination C. single displacement D. double displacement. 13 Which of the following is a balanced equation representing the decomposition of lead(IV) oxide? C. Pb2O → 2Pb + O A P6O2 → P6 + 2O B. PbO2 → Pb + O2 D PbO → Pb + O2 14 Boyle's law states that A the volume of a gas varies inversely with pressure B. the volume of a gas varies directly with pressure C. the temperature of a gas varies inversely with pressure D. the temperature of a gas varies directly with pressure 15. Charles's law states that A. the pressure of a gas is inversely proportional to its temperature in kelvins B. the volume of a gas is directly proportional to its temperature in kelvins C. the pressure of a gas is directly proportional to its temperature in kelvins D. the volume of a gas is inversely proportional to its temperature in kelvins 16 As the temperature of a fixed volume of a gas increases, the pressure will A. Vary inversely C. not change B. decrease D. increase 17. A chemical reaction is a process in which: A. products change into reactants C. substances change state B. the law of conservation of mass applies D. all of these 18 During a chemical reaction; A. new elements are produced C. atoms are destroyed B. atoms are rearranged D. elements are destroyed 19. An equation is balanced by: A. changing subscripts C . erasing elements as necessary

D. adding elements as necessary

B. adding coefficients

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Table of Contents Grade 10 Science Exam Ontario Answers

- 1. Understanding the eBook Grade 10 Science Exam Ontario Answers
 - The Rise of Digital Reading Grade 10 Science Exam Ontario Answers
 - Advantages of eBooks Over Traditional Books
- 2. Identifying Grade 10 Science Exam Ontario Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Grade 10 Science Exam Ontario Answers
 - User-Friendly Interface
- 4. Exploring eBook Recommendations from Grade 10 Science Exam Ontario Answers
 - Personalized Recommendations
 - Grade 10 Science Exam Ontario Answers User Reviews and Ratings
 - Grade 10 Science Exam Ontario Answers and Bestseller Lists

- 5. Accessing Grade 10 Science Exam Ontario Answers Free and Paid eBooks
 - Grade 10 Science Exam Ontario Answers Public Domain eBooks
 - Grade 10 Science Exam Ontario Answers eBook Subscription Services
 - Grade 10 Science Exam Ontario Answers Budget-Friendly Options
- 6. Navigating Grade 10 Science Exam Ontario Answers eBook Formats
 - o ePub, PDF, MOBI, and More
 - Grade 10 Science Exam Ontario Answers Compatibility with Devices
 - Grade 10 Science Exam Ontario Answers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Grade 10 Science Exam Ontario Answers
 - Highlighting and Note-Taking Grade 10 Science Exam Ontario Answers
 - Interactive Elements Grade 10 Science Exam Ontario Answers
- 8. Staying Engaged with Grade 10 Science Exam Ontario Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Grade 10 Science Exam Ontario Answers
- 9. Balancing eBooks and Physical Books Grade 10 Science Exam Ontario Answers
 - Benefits of a Digital Library
 - o Creating a Diverse Reading Collection Grade 10 Science Exam Ontario Answers
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Grade 10 Science Exam Ontario Answers
 - Setting Reading Goals Grade 10 Science Exam Ontario Answers
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Grade 10 Science Exam Ontario Answers
 - Fact-Checking eBook Content of Grade 10 Science Exam Ontario Answers
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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