

# Covalent Bonding

High School Chemistry Version

x10

## COVALENT BONDING

KEY IDEA

### COVALENT COMPOUNDS

COVALENT BOND

to hold together

COVALENT COMPOUND

formation that involves



HELDS TOGETHER BY THE  
ELECTROSTATIC ATTRACT

Overlapping of orbitals where  
electrons are shared

### How covalent bonding works



Chlorine has  
valence electrons.



Carbon has  
valence electrons.

and  
combine  
to form



This forms



Share  
one dot

## MOLECULES & Lewis structures

MOLECULE

LEWIS STRUCTURE



### Polar & Nonpolar Molecules

\_\_\_\_\_ have  
partial charges caused by  
unequal sharing of electrons.

\_\_\_\_\_ do not  
have partial charges.



### covalent properties

COVALENT PROPERTIES ARE MUCH MORE VARIOUS THAN IONIC



\_\_\_\_\_ between molecules so they are easier  
to separate.



\_\_\_\_\_ dissolve in water,  
which is also polar. \_\_\_\_\_ does not

# Visual Guided Notes

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**Annelies Wilder-Smith**



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