# Student Exploration: Limiting Reactants

Vocabulary: chemical equation, chemical formula, chemical reaction, coefficient, limitingreactant, molecule, product, reactant, subscript

Prior Knowledge Questions (Do these BEFORE using the Gizmo.)
Imagine you and your friends are making hot dogs. A complete hot dog consists of a wiener anda bun. At the store, you buy four packages of eight wieners and three bags of 10 buns.

- How many total hot dogs can you make? \_\_\_\_\_you can make 30 hotdogs with single bun and wieners \_\_\_\_\_
- Which ingredient limited the number of hot dogs you could make? the buns having 30 limited the number of hot dog i can make \_\_\_\_\_
- Which ingredient will you have leftovers of? i would have leftover of the wieners remaining 2 more.

### Gizmo Warm-up

Just as ingredients can be put together to make a new food, substances can combine during a **chemical reaction** to produce new substances. The substances that undergo change are called **reactants**. The new substances are **products**.

Sometimes during a chemical reaction, one type of reactant will be used up before the other reactants. This reactant is the **limiting reactant**. Using the *Limiting Reactants* Gizmo, you can determine which reactant is limiting in various scenarios.



To begin, make sure  $H_2 + O_2$  becomes H2O. The small "2" in  $H_2$ ,  $O_2$ , and  $H_2O$  is a subscript. Subscripts represent the number of atoms in a **molecule**.

- Use the sliders to set the number of O<sub>2</sub> molecules and H<sub>2</sub> molecules to two.
  - A. How many hydrogen molecules (H<sub>2</sub>) are there?\_\_\_\_1\_Hydrogen atoms?
  - B. How many oxygen molecules (O2) are there?\_\_\_1\_\_Oxygen atoms?\_\_2\_\_
- How many H₂O molecules do you think will form when these four molecules react?



**RC Schank** 

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