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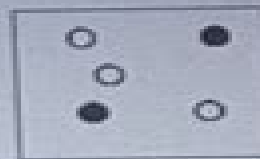
B



C



D



E

- ___ 1. Pure Element - only one type of atom present.
- ___ 2. Mixture of two elements - two types of uncombined atoms present.
- ___ 3. Pure compound - only one type of compound present.
- ___ 4. Mixture of two compounds - two types of compounds present.
- ___ 5. Mixture of a compound and an element.

Part 4: Read each description and determine whether it is a pure substance or mixture. Then further classify the matter (element, compound, homogeneous mixture, heterogeneous mixture)

Description	Pure Substance or Mixture?	Classification?
1. Chocolate syrup is added to milk and stirred	Mixture	Homogenous mixture (solution)
2. Copper metal (used to make wires)		
3. Sand is added to water		
4. Distilled water		
5. Tap water		
6. Helium gas (used to inflate a balloon)		
7. Table sugar		
8. Table sugar added to a cup of coffee and stirred		
9. The air we breathe		

Part 4: Polyatomic Ions

Elements And Compounds Section Review Answer Key

M Lipman



Elements And Compounds Section Review Answer Key:

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web assuming the volume of this mixture is 0.003 m³ kg of ceramic catalyst approximated as 1 kg of ceramic per 2 kg of solution thus 3 kg kg of ceramic the reactor volume is $V_r = 12 \text{ blocks} \times 5 \text{ kg ceramic block} = 0.003 \text{ m}^3 \text{ kg ceramic}$ 4h 0.72 m³
answer problem 13.12 continued 1 d the reactor must contain a hot solution of sodium

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